

ECB **IN VITRO** ANTIOXIDANT AND ANTIMICROBIAL POTENTIALS OF THREE EXTRACTS OF *AMARANTHUS HYBRIDUS* L. LEAF AND THEIR PHYTOCHEMICALS

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The study sought to determine the phytochemical components, antioxidant and antimicrobial activities of n-hexane, ethyl acetate and methanolic extracts of *Amaranthus hybridus* L. leaf. Three extracts of *A. hybridus* were examined for antimicrobial activity using disc diffusion assay. The different extracts demonstrated varied concentration-dependent antimicrobial activities against the test organisms. All extracts studied in this work were active against *E. coli*, *S. aureus*, *B. cereus*, *T. mentagrophyte* and *A. niger*. The methanol extract showed potent inhibitory activity against *T. mentagrophyte* when compared to a standard antifungal agent, fluconazole. *In vitro* antioxidant activities were studied spectrophotometrically using vitamin C as standard. There were significant correlations between the methanolic extract and vitamin C for 2,2-diphenyl-1-picrylhydrazyl (DPPH) scavenging, reducing ability, hydroxyl radical inhibitory and phosphomolybdate scavenging. The results of this study have shown that leaves of *A. hybridus* possess bioactive compounds which contributed to its antimicrobial and antioxidant properties.

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INTRODUCTION

Plants and plant-derived products have been a source of medicine for long. Even today, scientists and the general public recognize their value as a source of new and complementary medicines owing to their versatile applications.¹ Medicinal plants have been used for centuries as remedies for human diseases and offer a new source of biologically active chemical compounds as antimicrobial agents.² Plants have been known to contain or possess abundant phytochemicals, antimicrobials and pharmacologically active principles, which include anthraquinones, flavonoids, saponins, polyphenols, tannins and alkaloids.³

While orthodox medicine is generally accepted and preferred globally, the use of herbs and traditional medicines is often considered an equally acceptable alternative in many regions of the world.⁴ Traditional medicine is commonly used in developing countries where the cost of orthodox medicine and access to medical care are not available to a part of the population.⁴ The acceptance of traditional medicine as an alternative form of health care and the development of microbial resistance to the available antibiotics has led scientists to investigate the antimicrobial activity of medicinal plants.⁵ Likewise, the use of synthetic antioxidants are suspected to cause or promote negative health effects, hence stronger restrictions are being placed on their application and a trend to substitute them with naturally occurring antioxidants is developing.⁶ The role of medicinal plants in disease prevention or control has been attributed to antioxidant properties of their constituents.⁷

Keeping the above mentioned importance of medicinal plants in view, one of the medicinally important plants, *Amaranthus hybridus* L., also known as African spinach and 'terere' by most communities in Kenya, is cultivated in several regions of the world including South America, Africa, India, China, and the United States of America.⁸ In Kenya, its leaves are eaten like spinach or green vegetables. In Nigeria, *Amaranthus hybridus* leaves combined with seasonings are used to prepare soup.⁹ These leaves, when boiled and mixed with a groundnut sauce, are eaten as a salad in Mozambique or poured into a sauce and served over vegetables in West Africa.¹⁰ The plant is also used in the treatment of intestinal bleeding, excessive menstruation and diarrhea.¹¹

EXPERIMENTAL

Plant Material

The leaves of *Amaranthus hybridus* (Figure 1) were purchased from a market in Port Harcourt, Nigeria.



Figure 1. *Amaranthus hybridus* L. leaves.

The purchased leaves were identified and authenticated by Prof. (Mrs.) O. B. Green of the Department of Plant Science and Biotechnology, Rivers State University, Nigeria. The plant material was sorted out to obtain only fresh leaves which were washed with distilled water (without squeezing) to remove debris and dust particles. The washed leaves were air-dried for a few days under shade to prevent ultra-violet rays from altering the chemical constituents.^{2,12} The dry leaves were later pulverized using a manual blender.

Extraction

The dry pulverized leaves of *A. hybridus* (610 g) was macerated using 2.1 L of n-hexane in an aspiratory bottle at room temperature for 48 h with frequent stirring.¹³ Then the extract obtained was filtered into a conical flask using a funnel and a filter paper to obtain the n-hexane extract. The residue left was again subjected to second successive extraction with fresh n-hexane according to the procedure described above to obtain the second extract of n-hexane; this process was done 6 times to exhaustively extract the plant material. The same procedure was performed on the plant residue using 1.7 L of ethyl acetate and 1.2 L of methanol sequentially. The three extracts obtained were then separately concentrated using a rotary evaporator at 45 °C. The concentrated extracts were later weighed to obtain the yields and percentage yield for each extract was calculated.

Phytochemical Screening

Phytochemical examination was carried out on each of the extracts of *A. hybridus* leaf using standard methods. Each of the concentrated extract was subjected to qualitative tests via standard procedures to detect the presence of alkaloids, flavonoids, saponins, tannins, terpenoids, steroids and cardiac glycosides.^{14,15}

Phytochemical Quantification

The methods of Wenkam and Wills were adopted for the preparation and extraction of *A. hybridus* leaf for GC-MS analysis.^{16,17}

Antimicrobial Assay

Clinical isolates of two Gram-positive pathogenic bacteria (*Bacillus cereus* and *Staphylococcus aureus*), two Gram-negative pathogenic bacteria (*Escherichia coli* and *Pseudomonas aeruginosa*) and two pathogenic fungi (*Aspergillus niger* and *Trichophyton mentagrophyte*) were used for this study. The microbes were obtained from University of Port Harcourt teaching hospital, Nigeria.

Disc diffusion assay

Disc diffusion method of susceptibility testing, as described by Mahmodi *et al.* and Ndukwe *et al.* with some modifications was used to evaluate the antimicrobial properties of the extracts.^{18,19} In this method, broth cultures of microorganisms were first prepared by inoculating a colony or colonial material of each microorganism separately, into 10 mL sterile nutrient broth for the bacteria

and potato dextrose broth for the fungi. Inoculated nutrient broth tubes were incubated at 37 °C for 6 h to obtain a turbidity equivalent to 0.5 McFarland standards, while inoculated PDB broth tubes were incubated at 25 °C for 24 h. After incubation, the broth cultures of the bacteria were swab plated, separately, unto sterile nutrient agar plates. This was also done for the fungi using Sabouraud dextrose agar plates. The number of agar plates used corresponded to the number of dilutions of the investigated extracts, the number of microorganisms subjected to the testing, and the number of standard drugs (positive controls) used. Filter paper discs of 6 mm in diameter were impregnated separately with the *A. hybridus* leaf extracts and placed on the inoculated plates with the aid of sterile forceps. Antibiotic and antifungal impregnated discs were also placed on another set of inoculated plates to serve as positive controls. The plates were incubated at 37 °C for 24-48 h, after which the zones of inhibition around the discs were measured.

Antioxidant Assays

DPPH radical scavenging assay

The DPPH radical scavenging activity was determined according to the method reported by Sunil and Ignacimuthu with few modifications.²⁰ 1 mL of methanolic DPPH (0.15%) was mixed with 3 mL of each *A. hybridus* leaf extract at varying concentrations (0.25-2.5 mg L⁻¹) or vitamin C (reference antioxidant) and incubated in a dark room for 30 min. Thereafter, absorbance was measured at 515 nm. Scavenging activity of each extract and vitamin C was expressed as percentage and calculated using eqn. (1). Distilled water was used as blank.

$$\% \text{Scavenging activity} = \frac{A_c - A_s}{A_c} \times 100 \quad (1)$$

where A_c is absorbance of the control and A_s is absorbance of the sample.

Hydroxyl Radical Scavenging Assay

Hydroxyl radical scavenging activity was investigated using the method described by Bera *et al.* with few modifications.²¹ 1 mL of phosphate buffer (0.2 M, pH 7.2), 1 mL of test solution either *A. hybridus* leaf extract (0.25-2.5 mg L⁻¹) or vitamin C, 0.02 mL of ferric chloride (0.02 M) and 0.05 mL of phenanthroline (0.04 M) were introduced into a test tube. The reaction was triggered by adding 0.05 mL of 7 mM hydrogen peroxide. After 5 min of incubation at room temperature (25 °C), absorbance was measured at 560 nm using a UV spectrophotometer. Hydroxyl radical scavenging activity was expressed as percentage scavenging activity and calculated using eqn. (1). Methanol was used as blank.

Phosphomolybdate Assay

Free radical scavenging activity via the phosphomolybdate method was determined according to the method of Jayaprakash *et al.* as modified by Okoko and Diepreye.^{22,23} 0.2 mL of either *A. hybridus* leaf extract (0.25-

2.5 mg L⁻¹) or vitamin C (reference antioxidant) was mixed with 1 mL of phosphomolybdate reagent (4 mM ammonium molybdate, 28 mM sodium phosphate and 0.6 M sulphuric acid) and incubated in a water bath at 95 °C for 90 min. Absorbance was taken at 695 nm after allowing the content to cool. Free radical scavenging activity was expressed as percentage activity and calculated using eqn. (2). Distilled water was used as blank.

$$\% \text{ TAC} = \frac{A_c - A_s}{A_c} \times 100 \quad (2)$$

where TAC is total antioxidant capacity.

Reducing Ability

The ability of the extract to reduce Fe³⁺ was investigated according to Oyaizu method as modified by Okoko and Diepreye.^{24,23} *A. hybridus* leaf extract or vitamin C (0.5 mL) was mixed with 0.5 mL of phosphate buffer (0.2 M, pH 6.6) and 0.5 mL of potassium ferricyanide (1%) and incubated at 50 °C. After incubation for 20 min, 0.5 mL of trichloroacetic acid (10%) was added and centrifuged for 10 min at 3000 rpm. A portion of the upper layer (0.5 mL) was mixed with 0.5 mL distilled water and 0.1 mL ferric chloride (0.1%). After 10 min of incubation at room temperature, absorbance was measured at 700 nm. An increase in absorbance indicated greater reducing ability.

Statistical Analysis

Data obtained were expressed as means ± standard deviation (SD) of triplicates. All data were subjected to one-way analysis of variance (ANOVA) using SPSS (version 20) software. The values were considered to be significantly different when p<0.01. Means and standard deviations from the DPPH radical scavenging activity, hydroxyl radical inhibitory, reducing ability and phosphomolybdate assays are results of experiments performed in triplicate.

RESULTS AND DISCUSSION

Yields of extracts

Plants provide a large range of natural compounds belonging to different molecular families offering various medicinal properties.²⁵ Ethno-botanical information revealed that the plant selected in this study is traditionally used for various medicinal purposes.^{11,26} Extraction of *A. hybridus* leaf was carried out starting with a non-polar solvent (n-hexane), followed by a semi-polar solvent (ethyl acetate) and finally a polar solvent (methanol). Percentage yields of *A. hybridus* leaf extracts are given in Table 1. Percentage of extractable compounds varied from 2.5 to 3.9 %. This observation agrees with the work of Ibrahim *et al.* who reported the percentage yield of extracts to be in the order, methanol > ethyl acetate > n-hexane.²⁷ However, the yields of the extracts, as well as the bioactivity of the extracts prepared by maceration method has been reported to vary in

several studies.²⁸ It has also been suggested that the maceration method may be a better choice for the extraction of secondary metabolites.²⁸

Table 1. Extraction yield of *A. hybridus* leaf.

Extract	Weight of extract, g	Percentage yield, %
n-Hexane	15.51	2.50
Ethyl acetate	19.92	3.20
Methanol	24.31	3.90

Qualitative and quantitative assessment of phytochemicals

The qualitative phytochemical screening of *A. hybridus* leaf extracts revealed the presence of secondary metabolites (Table 2). Alkaloids, saponins, steroids, terpenoids, and glycosides were present in *A. hybridus* leaf, but flavonoids and tannins were absent. This result corroborates the findings of Maiyo *et al.*, who noted that leaf extract of *A. hybridus* contained steroids, terpenoids and cardiac glycosides but lacked tannins.²⁸ Cardiac glycosides are an important class of natural drugs that are widely used in the modern treatment of congestive heart failure and for the treatment of arterial fibrillation and flutter.²⁹ Terpenoids have been demonstrated to be active against bacteria, fungi, viruses and protozoa.^{30,31} Alkaloids have important pharmacological uses such as analgesics, antibacterial, antimalarial and anti-hypertensive.³²

Table 2. Phytochemical groups present in *A. hybridus* leaf.

Phytochemical group	n-Hexane extract	Ethyl acetate extract	Methanol extract
Alkaloids	+	+	+
Saponins	-	+	+
Tannins	-	-	-
Flavonoids	-	-	-
Steroids	+	+	+
Terpenoids	+	+	+
Cardiac glycosides	-	+	+

Key: + Present; - Absent

Table 3. Phytochemical percentage composition of *A. hybridus* leaf.

Phytochemical group	Percentage composition (%)
Alkaloids	0.004
Glycosides	0.009
Saponins	0.002
Flavonoids	Nil
Tannins	Nil
Total	0.015

GC-MS is preferred for more precise information in both qualitative analysis and quantitative determination.³³ A quantitative estimate of the percentage of alkaloids, glycosides and saponins components are given in Table 3. Glycosides (0.009 %) had the highest phytochemical presence in *A. hybridus* leaf, followed by alkaloids (0.004 %), with saponins (0.002 %) as the lowest. The medicinal value of the plant may be related to their constituent phytochemicals. According to Varadarajan *et al.*,

the phytochemicals and other chemical constituents of medicinal plants account for their medicinal value.³⁴ GC-MS analysis of methanolic extract of *A. hybridus* leaf indicated the presence of 20 compounds. The names and structures of identified phytochemicals with their retention time, molecular formula, molecular weight and abundance (peak area in %) are presented in Table 4. Among the identified phytochemicals, squalene is a compound that possesses antitumor, antioxidant, anticancer, antimicrobial, chemopreventive, pesticide and sun-screen properties; squalene has also been reported as an important precursor for the synthesis of phytosterols such as sitosterol, campesterol and stigmasterol. Capsaicin and dihydrocapsaicin are alkaloids and have been reported to have antioxidant activity, methyl commate B is a triterpene glycoside in nature. Triterpene glycosides are well known for their cytotoxic, antibacterial, antimicrobial, antiviral, insecticide, nematocidal, anticoagulant, hemolytic, antiparasitic, wound healing and antitumor activities. Quinazoline is an alkaloid having anticancer, antifungal and antibacterial properties while stigmasterol exhibits anti-cancer activities.³⁵⁻⁴⁰ Qualitative and quantitative studies have revealed that this plant is rich in phytochemical components and therefore could play important medicinal and physiological roles.

Antimicrobial activity of *A. hybridus* leaf extracts

The methanol extract of *A. hybridus* leaf demonstrated varied concentration-dependent antimicrobial activities against the test organisms (Table 5). The extract was active against *E. coli*, *S. aureus*, *B. cereus*, *A. niger* and *T. mentagrophyte*. However, there was no observable zone of inhibition against *P. aeruginosa*. Antimicrobial activities of the ethyl acetate extract of *A. hybridus* leaf are presented in

table 6. It was observed that the ethyl acetate extract demonstrated weak activity against *S. aureus*, *B. cereus*, and *E. coli* when compared with the positive control (streptomycin and ofloxacin). Only 50-200 mg mL⁻¹ of the extract had effect on *T. mentagrophyte* and *A. niger*, the other dilutions had no effect. There was no measurable zone of inhibition against *P. aeruginosa*. n-Hexane extract had little inhibitory effect on *S. aureus*, *E. coli* and *B. cereus*. There was no measurable inhibition zone against *P. aeruginosa*. However, the extract showed moderate antifungal activity in some of the concentrations compared to conventionally used fluconazole (Table 7). Several studies have indicated that Gram-negative bacteria are more resistant to antimicrobial agents than Gram-positive bacteria, due to the presence of a multilayered structure of Gram-

negative bacteria that is not present in Gram-positive bacteria.⁴¹ Results of this study showed that the extracts of *A. hybridus* leaf were active against *E. coli*, *S. aureus*, *B. cereus*, *T. mentagrophyte* and *A. niger* but were ineffective against *P. aeruginosa* (Tables 5-7). *P. aeruginosa* is considered one of the most rapidly growing bacteria in its resistance to existing antibiotics.^{19,42} The results of *A. hybridus* leaf indicate that methanol produced a more potent extract with increased antimicrobial activity, which inhibited greater number of bacterial strains and fungi. The observed antifungal potency of methanol extract can be attributed to two reasons. First, the nature of biologically active components (alkaloids, terpenoids and saponins) present.⁴³ Alkaloids, terpenoids and saponins, documented as plant metabolites, have been well known for their antimicrobial activities.⁴³ Secondly, the high polarity and strong extraction capacity of methanol may have given rise to a large number of active constituents responsible for the observed antifungal activity.

Table 4. Identified components of methanolic extract of *A. hybridus* leaf.

Compound	Molecular formula	Molecular weight	Retention time (min)	Peak Area (%)
4-(2,6,6-Trimethyl-1-cyclohexen-1-yl)-(E)- 3-buten-2-one	C ₁₃ H ₂₀ O	192	9.7024	0.30
Quinazoline	C ₈ H ₆ N ₂	130	10.441	0.61
N-(4-Methoxyphenyl)-2-hydroxyimino acetamide	C ₉ H ₁₀ N ₂ O ₃	194	11.487	0.57
Spiro[4.5] decan-6-one	C ₁₀ H ₁₆ O	152	12.072	1.10
Dihydrocapsaicin	C ₁₈ H ₂₉ NO ₃	307	16.771	0.14
Nonivamide	C ₁₇ H ₂₇ NO ₃	293	16.855	0.10
Capsaicin	C ₁₈ H ₂₇ NO ₃	305	17.131	9.70
N-(4-Chlorophenyl)-4-nitrophenyl ester carbamic acid	C ₁₃ H ₉ ClN ₂ O ₄	292	18.587	0.86
Squalene	C ₃₀ H ₅₀	410	19.977	0.94
cis-1,2-Dicarboximine-4-n-butyl cyclohexane	C ₁₂ H ₁₇ NO ₂	207	20.520	0.33
6-(4-Ethoxyphenyl)-3-methyl-4-oxo-4,5,6,7-tetrahydro-isopropyl ester-1H-Indole-2-carboxylic acid	C ₁₀ H ₁₁ NO ₃	193	20.901	0.33
2,7-Diphenylindole	C ₂₀ H ₁₅ N	269	21.931	0.24
N-[4-(Trifluoromethoxy)-phenyl]-1H-pyrazole-3-carboxamide	C ₁₁ H ₇ F ₃ N ₄ O ₄	316	22.033	6.45
Olean-12-ene	C ₃₀ H ₅₀	410	22.031	0.32
Lanosterol	C ₃₀ H ₅₀ O	426	22.977	1.97
N-Ethoxy-2-carbomethoxyaziridin	C ₁₅ H ₂₇ NO ₃	116	26.644	0.41
(3β,22E)-Chola-5,22-dien-3-ol	C ₂₄ H ₃₈ O	342	27.441	0.17
6-Methyl-1,5-diazabicyclo[3.1.0]hexane	C ₅ H ₁₀ N ₂	98	28.068	0.36
Stigmasterol	C ₂₉ H ₄₈ O	412	31.121	0.70
Methyl commate B	C ₃₁ H ₅₀ O ₃	470	31.352	0.07

Table 5. Antimicrobial activity of methanol extract of *A. hybridus* leaf.

MCO	Zone of inhibition (mm)						
	200 mg mL ⁻¹	100 mg mL ⁻¹	50 mg mL ⁻¹	25 mg mL ⁻¹	12.5 mg mL ⁻¹	6.25 mg mL ⁻¹	PCT
S.a	10.1±0.23	9.7±0.12	8.9±0.21	8.1±0.21	7.7±0.12	7.7±0.12	28.9±1.05(ST)
B.c	10.1±1.10	9.0±0.20	8.6±0.12	8.3±0.20	8.1±1.01	8.1±0.17	19.8±0.12(ST)
E.c	9.2±0.31	8.9±0.21	8.6±0.23	8.2±0.40	7.8±0.29	8.1±0.25	19.3±0.35(OF)
P.a	NA	NA	NA	NA	NA	NA	15.9±0.21(OF)
A.n	9.9±0.21	8.8±0.06	8.5±0.15	8.0±0.06	7.9±0.15	7.4±0.06	13.9±0.12(FZ)
T.m	9.8±0.23	9.2±0.21	8.8±0.31	8.3±0.25	7.4±0.25	NA	12.0±0.10(FZ)

NA - no action, MCO – microorganisms, B. c – *Bacillus cereus*, S. a – *Staphylococcus aureus*, E. c – *Escherichia coli*, P. a – *Pseudomonas aeruginosa*, A. n *Aspergillus niger*, T. m – *Trichophyton mentagrophyte*, PCT – positive control: ST- Streptomycin, OF – Ofloxacin, FZ – Fluconazole.

Table 6. Antimicrobial activity of ethyl acetate extract of *A. hybridus* leaf.

MCO	Zone of inhibition (mm)						
	200 mg mL ⁻¹	100 mg mL ⁻¹	50 mg mL ⁻¹	25 mg mL ⁻¹	12.5 mg mL ⁻¹	6.25 mg mL ⁻¹	PCT
S.a	9.1±0.27	8.5±0.15	NA	NA	NA	NA	28.9±1.05(ST)
B.c	9.3±0.47	9.1±0.21	8.5±0.23	8.4±0.25	8.0±0.21	7.3±0.21	19.8±0.12(ST)
E.c	8.9±0.06	7.6±0.17	NA	NA	NA	NA	19.3±0.35(OF)
P.a	NA	NA	NA	NA	NA	NA	15.9±0.21(OF)
A.n	9.7±0.27	8.2±0.12	7.7±0.23	NA	NA	NA	13.9±0.12(FZ)
T.m	9.8±0.12	9.1±0.21	8.6±0.12	NA	NA	NA	12.0±0.10(FZ)

NA - no action, MCO – microorganisms, B. c – *Bacillus cereus*, S. a – *Staphylococcus aureus*, E. c – *Escherichia coli*, P. a – *Pseudomonas aeruginosa*, A. n *Aspergillus niger*, T. m – *Trichophyton mentagrophyte*, PCT – positive control: ST- Streptomycin, OF – Ofloxacin, FZ – Fluconazole.

Table 7. Antimicrobial activity of n-hexane extract of *A. hybridus* leaf.

MCO	Zone of inhibition (mm)						
	200 mg mL ⁻¹	100 mg mL ⁻¹	50 mg mL ⁻¹	25 mg mL ⁻¹	12.5 mg mL ⁻¹	6.25 mg mL ⁻¹	PCT
S.a	9.5±0.36	8.8±0.21	NA	NA	NA	NA	28.9±1.05(ST)
B.c	9.4±0.59	9.1±0.23	8.8±0.15	8.2±0.25	7.7±0.61	7.6±0.29	19.8±0.12(ST)
E.c	8.9±0.59	8.8±0.23	8.4±0.15	NA	NA	NA	19.3±0.35(OF)
P.a	NA	NA	NA	NA	NA	NA	15.9±0.21(OF)
A.n	9.0±0.21	8.5±0.12	7.7±0.15	NA	NA	NA	13.9±0.12(FZ)
T.m	9.9±0.10	8.9±0.10	NA	NA	NA	NA	12.0±0.10(FZ)

NA - no action, MCO – microorganisms, B. c – *Bacillus cereus*, S. a – *Staphylococcus aureus*, E. c – *Escherichia coli*, P. a – *Pseudomonas aeruginosa*, A. n *Aspergillus niger*, T. m – *Trichophyton mentagrophyte*, PCT – positive control: ST- Streptomycin, OF – Ofloxacin, FZ – Fluconazole.

Table 8. ANOVA Result for antioxidant activity of *A. hybridus* leaf extracts.

Extract	Method			
	HRIA (560 nm)	DPPH (515 nm)	Phosphomolybdate (695 nm)	Reducing ability (700 nm)
Vitamin C	0.41±0.15	0.54±0.14	0.45±0.04	0.33±0.07
Methanol extract	0.43±0.06	0.51±0.13	0.44±0.04	0.29±0.04
n-Hexane extract	0.34±0.06	0.37±0.10*	0.36±0.04*	0.27±0.08
Ethyl acetate extract	0.41±0.18	0.49±0.08	0.40±0.04	0.28±0.05
P-value	0.1667	<0.0001	<0.0001	0.1685
F-value	1.742	12.37	11.36	1.761
Summary	Ns	S	S	Ns

Summary of the six varying concentrations (0.25, 0.50, 1, 1.5, 2.0 and 2.5 mg L⁻¹) of different solvent extract of *A. hybridus* leaf. Values are the mean ± SD of samples in triplicate. Means present a significant difference (p<0.01). *Significance difference versus vitamin C, S-significant, NS – No significant difference.

Although results of the antibacterial assay were not very promising, an increase in the concentration of the extracts may improve the inhibition. In addition, the antimicrobial activities displayed against *E. coli*, *S. aureus*, *B. cereus* T.

mentagrophyte and *A. niger* could be due to the synergistic effect of bioactive compounds such as squalene, quinazoline and methyl commate B, identified in the plant.

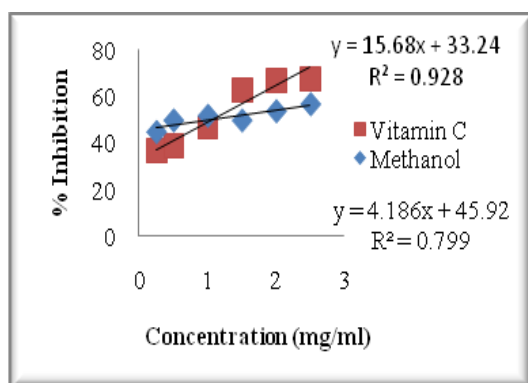


Figure 2. DPPH free radical scavenging activity of methanol extract of *A. hybridus* leaf.

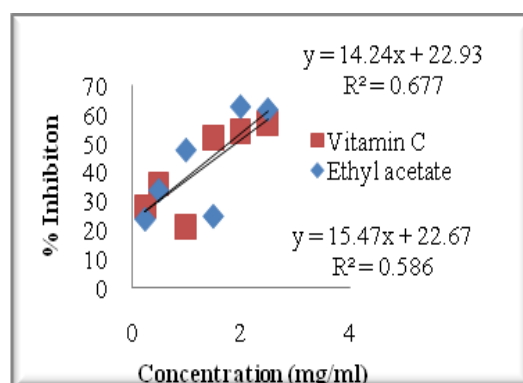


Figure 6. Hydroxyl radical inhibitory activity of ethyl acetate extract of *A. hybridus* leaf.

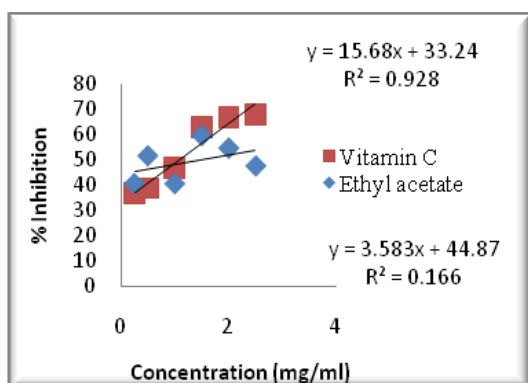


Figure 3. DPPH free radical scavenging activity of ethyl acetate extract of *A. hybridus* leaf.

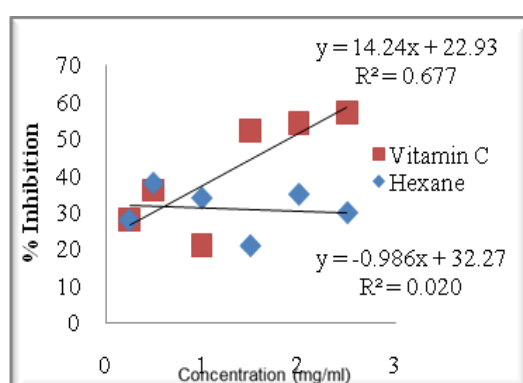


Figure 7. Hydroxyl radical inhibitory activity of n-hexane extract of *A. hybridus* leaf.

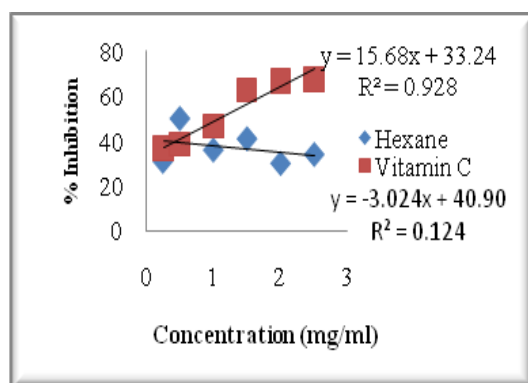


Figure 4. DPPH free radical scavenging activity of n-hexane extract of *A. hybridus* leaf.

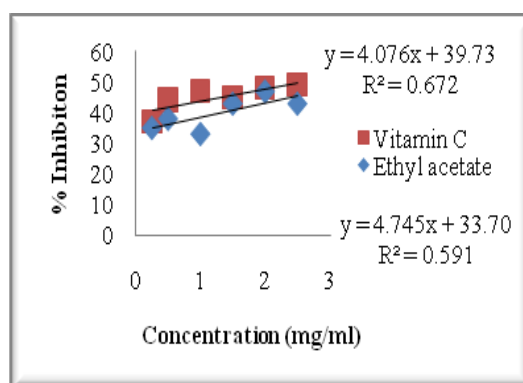


Figure 8. Phosphomolybdate assay of ethyl acetate extract of *A. hybridus* leaf.

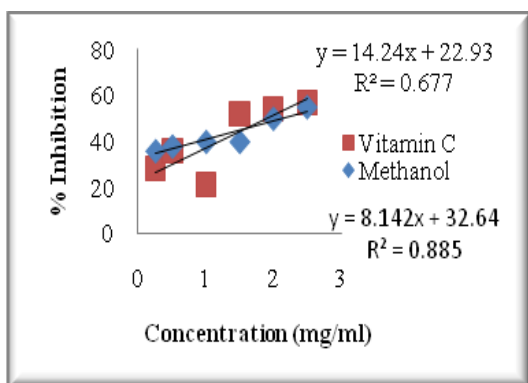


Figure 5. Hydroxyl radical inhibitory activity of methanol extract of *A. hybridus* leaf.

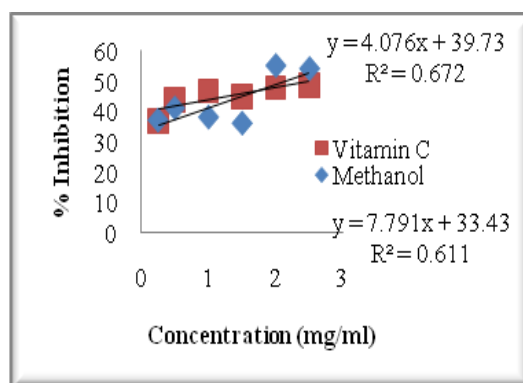


Figure 9. Phosphomolybdate assay of methanol extract of *A. hybridus* leaf.

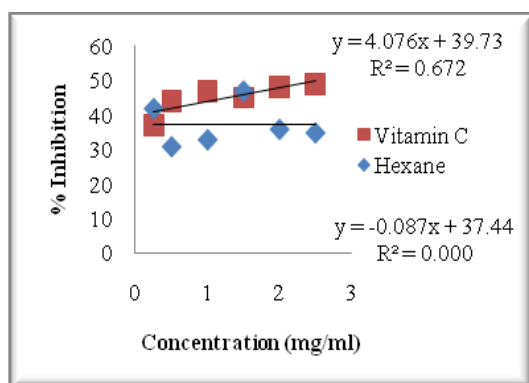


Figure 10. Phosphomolybdate assay of n-hexane extract of *A. hybridus* leaf.

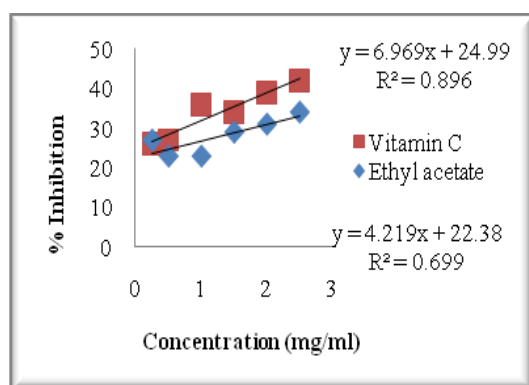


Figure 11. Reducing ability of ethyl acetate extract of *A. hybridus* leaf.

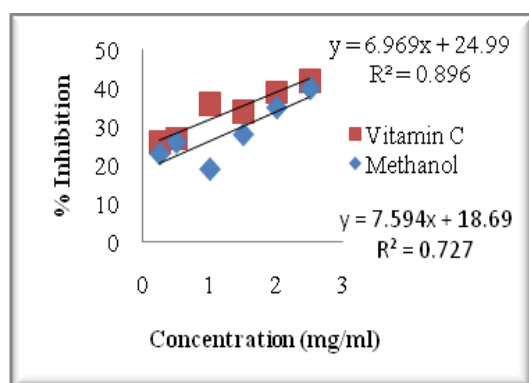


Figure 12. Reducing ability of methanol extract of *A. hybridus* leaf.

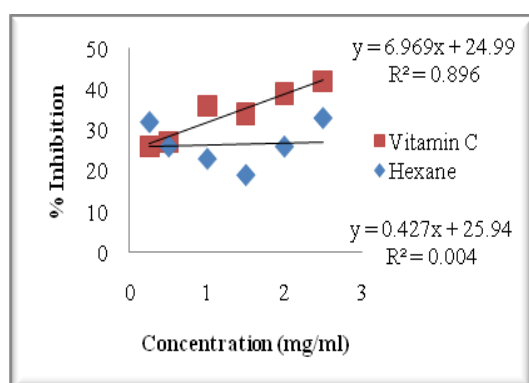


Figure 13. Reducing ability of n-hexane extract of *A. hybridus* leaf.

Antioxidant activity

Antioxidant activity of *A. hybridus* leaf was determined using DPPH radical scavenging, hydroxyl radical inhibitory, reducing power and phosphomolybdate assays. Previous researchers have reported that different methods for determining antioxidants activity give different results because of their different mechanistic principles.⁴⁴ DPPH assay is one of the best-known, accurate, and frequently employed methods for evaluating antioxidant activity.⁴⁵ It is a stable free radical because of delocalization of its unpaired electron over the whole molecule. Donation of protons to the DPPH radicals results to a corresponding change from violet to pale yellow in solution. Comparable scavenging activities of the plant extracts were observed with those of the standard compound (Table 8). The result revealed significant differences with n-hexane extract while for ethyl acetate and methanolic extracts, there were no significant differences. It was also observed that the scavenging effect of the plant extracts on DPPH radical was in the order, methanol > ethyl acetate > n-hexane extracts. Scavenging effect of *A. hybridus* leaf extracts could be attributed to the presence of stigmasterol, an anticancer phytosterol.^{35,40} It was observed that there exist a strong significant correlation ($R^2 = 0.7996$, Figure 2) in the activity of methanol extract of *A. hybridus* leaf when compared to vitamin C. In contrast, n-hexane extract showed a non-significant correlation with a negative slope ($y = -3.0247x + 40.907$, $R^2 = 0.1242$, Figure 4) indicating that there was little or no scavenging activity in the n-hexane extract. The trend in the behaviour of methanolic extract was similar to that of vitamin C. Figure 3 shows that the radical scavenging activity is in positive correlation with the concentrations of ethyl acetate extract used. However, correlation in the case of n-hexane was found to be non-significant (Figure 4). This result is consistent with another report of a strong correlation of antioxidant activity in methanolic extract.⁴⁶

Reducing power assay is often used to assess the ability of an antioxidant to donate an electron.²⁴ In this study, the ability of extracts was estimated in order to determine the potential of the extracts to reduce Fe^{3+} to Fe^{2+} by electron donation. This is associated with the presence of a redundant or complex molecule serving as an electron donor and/or free radical scavengers. In this study, the reducing power of the plant extracts were found in the order, methanol > ethyl acetate > n-hexane (Table 8), indicating that the methanolic extract has a better reducing power than other extracts. Methanolic extract of *A. hybridus* leaf may, therefore, contain high amount of reductants compared to n-hexane and ethyl acetate extracts. Therefore, methanolic extract can act as electron donor and could react with free radicals to convert them into more stable products and terminate the free radical chain reactions. This result confirms the findings of El-Hashasa *et al.*, who reported that the reducing power of a plant correlates with its phenolic content.⁴⁷ Figure 12 depicts that the reducing ability is in positive correlation with the concentrations of the extract (methanol) used. Significant positive correlations (Figure 11) were observed between the ethyl acetate extracts and vitamin C. While the n-hexane extract exhibited a non-significant correlation (Figure 13).

The hydroxyl radical is one of the most reactive oxygen species in living systems, which can react with all possible molecules in living organisms, especially proteins, DNA and

lipids.⁴⁸ Thus, removing OH radical is very important for the protection of biological systems. In this study, the methanol extract of *A. hybridus* leaf was more active than other extracts and was slightly better than vitamin C (Table 8). Correlation coefficient of methanol extract is $R^2 = 0.885$, showing that the inhibitory activity of the plant was slightly higher than vitamin C ($R^2 = 0.6773$, Figure 5). Ethyl acetate extract of *A. hybridus* leaf scavenged hydroxyl radical at concentrations of 2.0 mg mL^{-1} to 2.5 mg mL^{-1} which may be attributed to the presence of capsaicin. Capsaicinoids with phenolic structure can donate hydrogen to hydroxyl radical, thereby neutralizing it into water. The R-value of ethyl acetate appeared to be better than that of n-hexane (Figure 6 and 7). This could be as a result of the polarity index of n-hexane (0.1). There was a strong linear relationship between the activities of ethyl acetate and vitamin C (Figure 6). In contrast, n-hexane exhibited no correlation or relationship between the two variables (Figure 7). Alkaloids are major antioxidants in natural products, and their antioxidant activities have been proven in recent studies.^{49,50} The activity of these plant extracts, especially methanol and ethyl acetate extracts are indications of the presence of phytochemicals that could be responsible for the observed antioxidant activity.

Phosphomolybdate test measures the ability of an extract to destroy free radicals by transferring an electron to the later. The antioxidants present in the extract reduced molybdate (VI) to molybdate (V) and this was measured spectrophotometrically at 695 nm.²² Comparable scavenging activities of the plant extracts were observed with those of the standard compound (Table 8). The result revealed significant differences with n-hexane extract while for ethyl acetate and methanolic extracts, there were no significant differences. However, the antioxidant activity of vitamin C, an antioxidant used as the positive control, had a significantly better correlation than *A. hybridus* leaf (Figures 8-10). The observed differences in antioxidant properties may be due to the polarity of extraction solvents.

CONCLUSION

Methanolic extract of *A. hybridus* leaf showed significant antioxidant property. The leaf of *A. hybridus* is rich in various phytochemicals and had moderate to significant antifungal property but showed quite weak antibacterial activity. Bioactive compounds present in this leaf can be harnessed as natural antioxidants and pharmaceutical products.

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