



DEVELOPMENT OF NEW ENTITIES OF 1,3,4-THIADIAZOLES FOR ANTIOXIDANT AND ANTIMICROBIAL ACTIVITIES.

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Abstract:

1, 3, 4- Thiadiazole is one of the heterocyclic compounds having different versatile activities such as anti-diabetic, anti-microbial, anti-convulsant, anti-inflammatory and anti-cancer. A new 1,3,4-Thiadiazole derivatives with Schiff base compound synthesized. The starting material N-(4-chlorophenyl)hydrazine carbothiamide synthesized by the reaction between aniline derivatives and carbondisulfide, hydrazine hydrate followed by synthesis of 5-(4-aminophenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazole-2-amine. The chemical structures were confirmed by the elemental analysis IR, UV and H¹ NMR and Thin layer chromatography and perform pharmaceutical activity carried out against S aureus, E Coli, K. Pneumonia, M luteus and candida albicans and compared with the standards from which 4-NO₂, 4-Cl, 2-OH,-H derivatives shown promising antibacterial activity and 4-Cl derivative shown promising anti-fungal activity.

Keywords: 1, 3, 4- Thiadiazole, anti-bacterial activity, 4-Cl derivative, E Coli, Schiff base.

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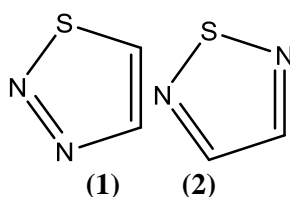
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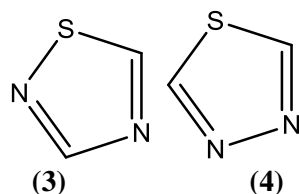
INTRODUCTION

Microbial infections are caused by many types of bacteria, viruses, and fungi, are one of the most common illnesses that kill millions of people throughout the world. The actions of microbes have a significant impact on humans. Microorganisms play a larger role in our lives than most of us realise. They have shaped our current environment, and their actions will have a significant impact on our future. Microorganisms should not be thought of as distinct from humans, but rather as an integral component of our lives. They are used in the production of dairy products, some meals, certain pharmaceuticals and therapeutic agents, as well as the production of other compounds.

Heterocyclic structures are constantly present in organic chemistry research and development. There are millions of heterocyclic structures with unique characteristics and biological significance. An acceptable synthetic approach was used to make a variety of thiadiazoles, which were then characterised using elemental analysis and spectrum data[1].

The ring contains multiple isomers of Thiadiazole: 1,2,3 Thiadiazole (1), 1,2,5 Thiadiazole (2), 1,2,4 Thiadiazole (3), 1,3,4 Thiadiazole (4), 1,3,4 Thiadiazole (5), 1,3,4 Thiadiazole (6), 1,3,4 Thiadiazole (7), 1,3,4 Thiadiazole (4)





"Fischer" initially described 1,3,4-thiadiazole in 1882, and "Busch" and his colleagues further improved it. Thiadiazole has a five-membered ring structure with the chemical formula $C_2H_2N_2S$, which comprises two carbon atoms, two hydrogen atoms, two nitrogen atoms, and one sulphur atom. A five-membered ring structure with two or more heteroatoms, one of which being nitrogen, is designated by the ending azole. Thiadiazoles are linked to a wide range of biological activities, most likely as a result of their $-N=C-S-$ The heterocyclic nucleus of the 1,3,4-thiadiazole moiety contains sulphur at position -1 and two nitrogen atoms at positions 3 and 4. In the realm of medicinal chemistry, thiadiazole derivatives have a distinctive position. Thiadiazole's Sulphur atom improves lipid solubility and is meso ionic in nature. Pharmaceuticals, cyanide dyes, oxidation inhibitors and metal complexing agents are just a few of the uses for the 1,3,4-thiadiazole ring system. Thiadiazole is a multifunctional moiety with a wide range of biological activity. Acetazolamide, Methazolamide, Sulfamethazole, and other medicines containing the thiadiazole nucleus are available on the market. Thiadiazole can be used to substitute the thiazole molecule bio-isosterically. As a result, it behaves similarly to third and fourth generation cephalosporins and can be employed in antibiotic formulations. Thiadiazole derivatives have intriguing biological activity, which is likely due to the strong aromaticity of this ring structure, which results in excellent in vivo stability and low toxicity in higher vertebrates, including humans.

EXPERIMENTAL METHODOLOGY

N-(4-chlorophenyl) hydrazine carbothioamide is synthesised from 4-chloroaniline. Cyclization of N-(4-chlorophenyl) hydrazine carbothioamide with para amino benzoic acid in the presence of concentrated sulfuric acid is employed as a reactant in the synthesis of 5-(4-aminophenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine. After being treated with different substituted Aldehydes, it is further transformed to imines. The Staudinger

Imines reaction of Schiff bases was used to synthesise the final compounds.

STEP-1: Synthesis of N-(4-chlorophenyl) hydrazine carbothioamide:

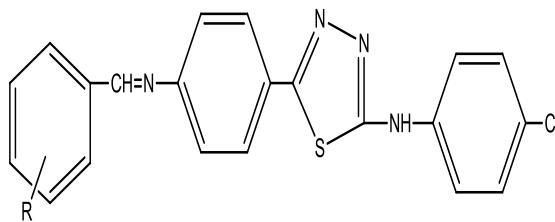
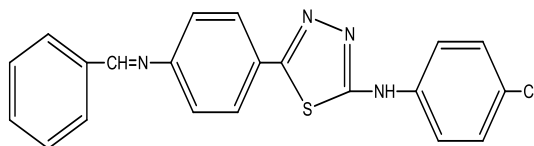
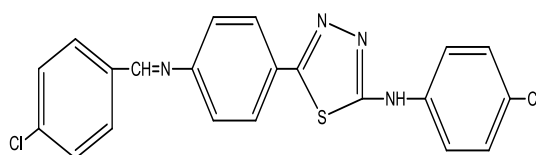
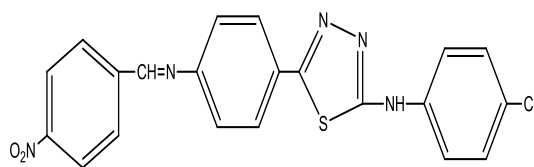
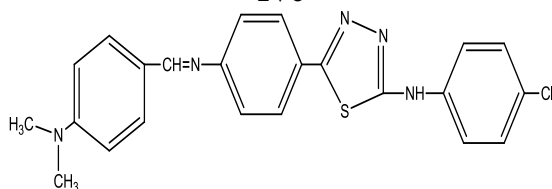
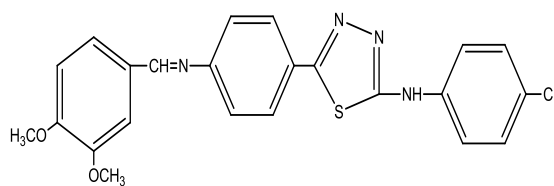
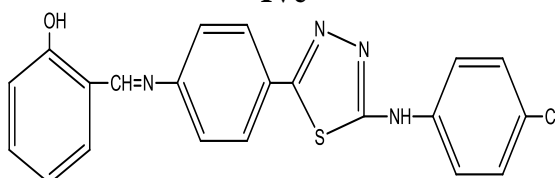
In a round bottom flask placed equimolar quantity of para chloro aniline (0.1mole), potassium hydroxide (0.1mole) and methanol (20ml) stirred on magnetic stirrer, maintained the temperature below $10^\circ C$ and carbon disulphide (0.1mole) added drop by drop, stirring continued for 2-3hrs at $0^\circ C$ and to it hydrazine hydrate (0.2mole) added drop by drop, stirring was continued at $0^\circ C$ for 1hr, then reflux the reaction mixture for 3-4hrs. The reaction mixture was poured into ice cold water then precipitation was formed. The product was collected by filtration, dried and purified with ethanol.

STEP-2: Synthesis of 5-(4-aminophenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine:

In a round bottom flask equimolar quantity of N-(4-chlorophenyl) hydrazine carbothioamide (0.1mole) and para amino benzoic acid(0.1mole) and sulfuric acid(4ml) in 20ml of methanol containing 2-3drops of glacial acetic acid were refluxed for 4-6hrs. The reaction mixture was poured in ice cold water and add 5% $NaHCO_3$ and the product was collected by filtration, dried and purified with methanol, crystalline product was obtained.

STEP-3: Synthesis of 5-(4-(benzylidene amino) phenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine:

In a round bottom flask placed 5-(4-aminophenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine (0.01mole) and benzaldehyde (0.01mole) containing 2-3 drops of glacial acetic acid and 20ml of ethanol were refluxed for 6 hrs at $20^\circ C$. The reaction monitored by TLC (7:3 ethylacetate: Hexane). Then, the reaction mixture was cooled to room temperature and poured in to crushed ice. The product was separated by filtration and dried. It was purified recrystallise from methanol, crystalline product was obtained.

Compound III**IV****IVa****IVb****IVc****IVd****IVe****IVf****Physical data and spectral data of synthesised compounds:**

S.no.	Compound	R	Mol. Formula	Mol. Weight	Melting point	% Yield
1	III	Intermediate	C ₁₄ H ₁₁ ClN ₄ S	302.78	120°C	82%
2	IV-a	H	C ₂₁ H ₁₅ ClN ₄ S	390.89	110°C	68.7%
3	IV-b	4-Cl	C ₂₁ H ₁₄ Cl ₂ N ₄ S	425.33	115°C	73.5%

4	IV-c	4NO ₂	C ₂₁ H ₁₄ ClN ₅ O ₂ S	435.85	142°C	56.9%
5	IV-d	4-N, N- Dimethyl	C ₂₃ H ₂₀ ClN ₅ S	433.96	172°C	44.5%
6	IV-e	3,4-Di-methoxy	C ₂₃ H ₁₉ ClN ₄ O ₂ S	450.94	138°C	49.8%
7	IV-f	2-OH	C ₂₁ H ₁₅ ClN ₄ OS	406.89	85°C	67.8%

Spectral data of 5-(4-aminophenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine (III): IR (KBr) cm⁻¹: 3422.59(-NH₂), 3220.81(NH), 2984(Aromatic C-H), 1599 and 1492(C=C), 1039(C-N), 1634(C=N), 772.63(C-Cl). ¹H-NMR (300 MHz, DMSO) δ ppm: ¹H-NMR (300 MHz, DMSO) δ ppm: 8.4(s, -1H, NH), 7.2– 8.2(m,8H, Ar-H), 4.65(s,2H, NH₂). EI-MS: M+2 peak observed at 304.

Spectral data of 5-(4-(Benzylidene amino)phenyl)-N-(4-Chlorophenyl) -1,3,4-Thiadiazol-2-Amine (IV): IR (KBr) cm⁻¹: 3283.81(NH), 2984(Aromatic C-H), 1634(C=N), 784.42(C-Cl), ¹H-NMR (300 MHz, DMSO) δ ppm: 8.45(s, -1H, NH), 7.40– 8.18(m,13H, Ar-H), 4.39(d,2H, CH). EI-MS: M+2 peak observed at 437.

Spectral data of (E)-5-(4-((4-chlorobenzylidene)amino)phenyl)-N-(4-chlorophenyl)-1,3,4-thiadiazol-2-amine (IV-b) : IR (KBr) cm⁻¹: 3422.59(-NH₂), 3220.81(NH), 2984(Aromatic C-H), 1599 and 1492(C=C), 1039(C-N), 1634(C=N), 772.63(C-Cl). ¹H-NMR (300 MHz, DMSO) δ ppm: ¹H-NMR (300 MHz, DMSO) δ ppm: EI-MS: M+2 peak observed at 425.

Spectral data of N-(4-chlorophenyl)-5-(4-((4-nitrobenzylidene)amino)phenyl)-1,3,4-thiadiazol-2-amine (IV-C) : IR (KBr) cm⁻¹: 3283.81(NH), 2984(Aromatic C-H), 1634(C=N), 784.42(C-Cl), 1358 & 1537(C-NO₂). ¹H-NMR (300 MHz, DMSO) δ ppm: 8.45(s, -1H, NH), 7.40– 8.18(m,13H, Ar-H), 4.39(d,2H, CH). EI-MS: M+2 peak observed at 392.

Spectral data of N-(4-chlorophenyl)-5-(4-((4-dimethylamino)benzylidene)amino)phenyl)-1,3,4-thiadiazol-2-amine (IV-d) : IR (KBr) cm⁻¹: 3283.81(NH), 2984(Aromatic C-H), 1634(C=N), 784.42(C-Cl), 1335-1250(C-N). ¹H-NMR (300 MHz, DMSO) δ ppm: 8.45(s, -1H, NH), 7.40– 8.18(m,13H, Ar-H), 4.39(d,2H, CH). EI-MS: M+2 peak observed at 435.

Spectral data of N-(4-chlorophenyl)-5-(4-((3,4-dimethoxybenzylidene)amino)phenyl)-1,3,4-thiadiazol-2-amine (IV-e) : IR (KBr) cm⁻¹: 3283.81(NH), 2984(Aromatic C-H), 2835(C-H of OCH₃), 1634(C=N), 784.42(C-Cl). ¹H-NMR (300

MHZ, DMSO) δ ppm: 8.45(s, -1H, NH), 7.40– 8.18(m,13H, Ar-H), 4.39(d,2H, CH). EI-MS: M+2 peak observed at 452.

Spectra data of 2-(((4-(5-((4-chlorophenyl)amino)-1,3,4-thiadiazol-2-yl)phenyl)imino)methyl)phenol (IV-f) : IR (KBr) cm⁻¹: 3283.81(NH), 2984(Aromatic C-H), 3345(C-OH), 1634(C=N), 784.42(C-Cl). ¹H-NMR (300 MHz, DMSO) δ ppm: 8.45(s, -1H, NH), 7.40– 8.18(m,13H, Ar-H), 4.39(d,2H, CH). EI-MS: M+2 peak observed at 408.

PHARMACOLOGICAL EVALUATION OF SYNTHESIZED COMPOUNDS:

Antibacterial activity: By the Agar Well Diffusion technique, roughly 25ml of nutritious agar medium was sterilised (autoclaved at 120°C for 20 minutes) and put into sterile Petri dishes, which were set aside for solidification. Bacterial lawns were created by using a series L-shaped glass rod to disseminate the bacterial suspension (1ml/100ml, or cu/ml) across the surface of the agar plates. Using a sterile borer, wells were punched in the agar plates (6mm). The prepared concentrations (100g/ml, 500g/ml, 1000g/ml) of the test sample, as well as the control and standard, were aseptically placed into 50l wells. To allow optimal diffusion of the solution into the nutrient agar medium, the plates were left undisturbed in the refrigerator for at least 2 hours. The plates were then incubated at 37°C for 24 hours. After the incubation time, the diameter of each well's zone of inhibition was manually measured and tabulated using a millimetre scale. Whole experiment was carried out in triplicate. Simultaneously dimethyl sulfoxide was maintained as control to observe the solvent effect on bacteria. Compounds IVa, IVf shows 7mm zone of inhibition against *Klebsiella pneumoniae* at 1000µg/ml the standard drug shows 10mm, in case of *Micrococcus luteus* compounds IVa, IVb, IVc shows 8,9,8mm zone of inhibition respectively, the standard drug ciprofloxacin shows 14mm at concentration of 1000µg/ml, in Case of *Staphylococcus aureus* compound IVb shows 3mm but the standard ciprofloxacin shows 6mm of zone of inhibition. In case of *E coli* compound IVb shows 2mm but the standard shows 3mm zone of inhibition.

Antifungal activity: By agar well diffusion method. 1,3,4 thiadiazole derivatives having less active towards the fungi. In case of *Candida albicans* the compounds IVa, IVc, IVd shows 4mm zone of inhibition but the standard drug shows 14 mm at the concentration of 1000µg/ml.

Anti-oxidant activity: By using the free radical scavenging property of DPPH, The IC₅₀ values of all synthesized test compounds were found between 11.3µM to 96.34µM. All the compounds were tested at 1nM to 1mM concentrations and results were compared with standard drug (Ascorbic acid) at the same concentrations. Compound IV-a (R= 5-H) & IV-b (R= 4-Cl) showed (21.6µM and 22.8µM) effective antioxidant activity. And IV-e (R= 4-N, N-Dimethyl) comparatively weak (96.34µM) antioxidant activity.

CONCLUSION:

New 1,3,4-Thiadiazole derivatives showed promising antimicrobial activity. Compounds **R= 4-Cl**, **R= 4-NO₂**, **R=2-OH**, **R= H** were found to be the more potent antibacterial activity respectively towards gram positive (*Micrococcus luteus*) and gram negative (*klebsiella pneumonia*) and compound **R= 4-Cl** found to be more potent antifungal activity towards *Candida albicans* and *Aspergillus Niger*. New 1,3,4-Thiadiazole derivatives showed promising antioxidant activity. Compounds IV-a (**R=H**)& IV-b (**R= 4-Cl**) were found to be more potent antioxidant compounds among the all test compounds.

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CONFLICT OF INTEREST

All the authors declare that they have no conflict of interest.

ABBREVIATION USED

All the abbreviations used have been described in the text.

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