



COMPUTATIONAL INVESTIGATION OF NORFLOXACIN DERIVATIVES PROPERTIES RELATED TO THEIR BIOLOGICAL ACTIVITY

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Abstract: Background: Fluoroquinolones have great efficacy, broad spectrum of activity, used for treatment of several infections, due to miss use and uncontrolled use of antibacterial agents, lead to emergence of bacterial resistance. Therefore new fluoroquinolones derivatives were synthesized and their biological activities were studied according to their some computational properties. **Aim:** To predict the antibacterial activity of newly synthesized norfloxacin derivatives depending on their computational investigated properties. **Material and method:** Seven norfloxacin derivatives have been designed; their physical properties have been investigated using quantum chemistry computational methods. **Results:** There were six descriptions (one electronic, three physicochemical and two geometric) have highly compatible with Norfloxacin derivative activity, can be used to predicts of their activity by using QSAR equation of Norfloxacin derivatives.

$$pMIC = 0.0795 + 6.12237 \times 10^{-3}(\eta) - 2.347 \times 10^{-3}(B.P) - 65.378 \times 10^{-3}(\log P) + 0.752(MR) + 6.000995 \times 10^{-3} \times (NI - C - NAA) - 3.92216 \times 10^{-3}(NI - C - SA)$$

In the above equation, we take the following notations for simplicity:

MR: Molecular Refractivity

NI: N Imine

NAA: N Amine angle

SA: S angle

Conclusion: QSAR would lead to robust equation and predictive models are capable of making accurate and reliable calculations for a new Norfloxacin analog

Keywords: QSARs, DFT, Molecular Properties, GAMESS, Norfloxacin, Antibacterial resistant.

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INTRODUCTION

Fluoroquinolones are a group of antimicrobial agents used worldwide as they have great efficacy, broad spectrum of activity, and safety [1]. They are used for treatment of several infections as urinary tract infection [2], respiratory tract infection [3], and skin infection [4]. Norfloxacin is one of the second generation fluoroquinolones [5], which act through inhibition of topoisomerase II or gyrase enzyme in gram negative bacteria, while it inhibits topoisomerase IV in gram positive bacteria [6]. In addition to its antibacterial action, it has anti-inflammatory effect through its ability to reduce the cytokines level and white blood cells aggregation [7]. Due to miss use and uncontrolled use of antibacterial agents, lead to emergence of bacterial resistance, therefore,

novel antibacterial agents or modification of the per-existing agents is necessary to overcome the bacterial resistance [8].

Quantitative Structure Activity Relation-Ship (QSAR)

The modern QSAR began in the early 1960s, however, as long ago as 1816 scientists were making predictions about physical and chemical properties. The first investigations into the correlation of biological activities with physicochemical properties such as molecular weight and aqueous solubility began in 1841. Throughout the 20th century QSAR progressed, though there were many lean years [9-10]. In 1962 came the seminal work of Corwin Hansch and co-workers, predicts biological activities $\log(I/C) = 0.94 \log(P) + 0.87$, initially that interest lay largely within medicinal chemistry and drug design [11].

METHODOLOGY

Computational Methods

A series of Norfloxacin derivatives tested for their activities were selected for the present study and the program of Windows Chem SW was adopted for molecular modeling studies. The molecules were generated and energy minimization was carried out by using Molecular Modeling Program, all calculations are carried out by General Atomic and Molecular Electronic Structure System (GAMESS) software [12-14]. After minimized energy, physicochemical properties were calculated for all studied molecules and the results are shown in table (2-1).

Selected Norfloxacin derivatives [15]

Norfloxacin is one of the second generation fluoroquinolones, it act through inhibition of bacterial topoisomerase enzyme by that prevent bacterial DNA replication, leading to bactericidal effect. Due to the uncontrolled use and miss use of the fluoroquinolones lead to the emergence of bacterial resistance so; it became necessary to synthesize of new derivative to overcome

bacterial resistance. It was found that the incorporation of bulk chemical group at carbon 7 of the piperazine ring lead to reduction in the affinity toward the bacterial efflux pump, which considered as one of the major routs of bacterial resistance. Depending on this background, six norfloxacin derivatives were synthesized through incorporation of triazole derivatives at C-7 piperazine ring.

Table 2-1: Activity of Norfloxacin Derivatives

Compound	IUPAC name	MIC (mcg/ml)
III	7-(4-(2-aminothiazol-4-yl) piperazin-1-yl)-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid	6.8
IVa	(E)-1-ethyl-6-fluoro-7-(4-(2-((4 hydroxybenzylidene)amino)thiazol-4-yl)piperazin-1-yl)-4-oxo-1, 4-dihydroquinoline-3-carboxylic acid	8.4
IVb	(E)-1-ethyl-6-fluoro-7-(4-(2-((4-methoxybenzylidene)amino)thiazol-4-yl)piperazin-1-yl)-4-oxo-1, 4-dihydroquinoline-3-carboxylic acid	8.6
IVc	(E)-7-(4-(2-((4-(dimethylamino)benzylidene)amino)thiazol-4-yl)piperazin-1-yl)-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid	8.8
IVd	(E)-7-(4-(2-((4-chlorobenzylidene)amino)thiazol-4-yl)piperazin-1-yl)-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid	8.7
IVe	(E)-7-(4-(2-((4-bromobenzylidene)amino)thiazol-4-yl)piperazin-1-yl)-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid	9.4
IVf	(E)-1-ethyl-6-fluoro-7-(4-(2-((4-methylbenzylidene)amino)thiazol-4-yl)piperazin-1-yl)-4-oxo-1, 4-dihydroquinoline-3-carboxylic acid	8.3

Results

Computer Calculations: The calculated description of electronic structures and physiochemical properties of properties for Norfloxacin was: Geometric structures, shown in table (1).

Table (1-a): Norfloxacin Derivatives Calculated Descriptors by GAMSS Program

Compound	MIC (µg/ml) or (ppm)	pM IC	HO MO = IP (ev)	LU MO = A ev	Π=L UM O-HO MO (ev)	μ=(HOM O+LUM O)/2 (ev)	ω= μ ² /2 Π (ev)	more(+) ev charge atom	more(-) ev charge atom	Kinetic Energy (Kcal /Mol)	Dipole (Debye)	Potential Energy (Kcal /Mol)	Boiling Point (k)	ΔH _f (kJ/mol)
III	68	-1.835	-7.201	-2.509	4.692	-4.855	2.512	0.781	-0.777	1074381.56	11.643	-2155836.23	870.618	-203.15
IVf	83	-1.919	-8.544	-2.765	5.779	-5.654	2.766	0.6224	-0.8343	1266193.	11.141	-2541501.1	954.743	-223.06
IVa	84	-1.924	-8.533	-2.474	6.059	-5.5035	2.4995	0.6702	-0.8683	1288636.293	10.597	-2586343.4	977.98	-368.26
IVb	86	-1.934	-8.541	-2.711	5.83	-5.626	2.715	0.5316	-0.7657	1326017.805	12.5887	-2655334.03	966.797	-355.28
IVd	87	-1.939	-8.485	-2.14	6.34	-5.31	2.224	0.7004	-0.8559	1528154.594	7.956949	-3065937.453	961.839	-218.16
IVc	88	-1.944	-8.552	-2.489	6.063	-5.5205	2.524	0.7356	-0.8745	1325225.816	13.40779	-2659546.432	977.625	-176.17

IVe	94	-1.9731	-8.567	-2.53	6.037	-5.5485	2.5498	0.703771	-0.838836	2853455.51		-5711347.338	973.845	-176.09
	dependent variable variation	R ²	0.942	0.011	0.916	0.711	0.113	0.135	0.27	0.41	0.056	0.41	0.867	0.0001
		Slope	0.074	-0.02	-0.069	0.118	0.0001	0.199	0.53	-5.00E-08	0.005	2.00E-08	-0.001	-2E-06

Table (1-b): Norfloxacin Derivatives Calculated Descriptors by GAMSS Program

Compound	MIC (µg/ml or ppm)	Log P	Accessible Area (Å ²)	Molecular Area (Å ²)	Solvent Excluded Volume (Å ³)	Number of H Bond Acceptors	Number of H Bond Donors	Mol Refractivity	Partition Coefficient	Entropy cal/(mol K)	Log S	PKa, Atom : 13	Balaban Index	Molecular Topological Index
III	68	3.45	642.57	349.302	318.129	8	2	10.867	2.174	154.964	5.3543	2.628	849776	15852
IVf	83	5.522	836.47	465.716	429.147	8	2	14.561	0.637	0	7.2316	2.6281	2557793	36739
IVa	84	4.645	818.129	454.988	419.511	9	3	14.25	0.372	0	6.80543	2.628	2557793	35867
IVb	86	4.908	853.622	473.427	434.998	9	2	14.714	0.4825	0	6.9248	2.62809	293891	39391
IVd	87	5.593	830.973	462.337	427.497	8	2	14.582	0.85146	0	9.999	0	255779	35431
IVc	88	5.319	880.878	493.371	456.36	9	2	15.3931	0.821958	0	9.999	0	3350863	43371
IVe	94	5.863	838.67	467.34	432.682	8	2	14.8738	1.0015	0	7.67904	2.62809	255779	35431
	dependent variable variation	R ²	0.8331	0.834	0.846	0.05	0.0001	0.884	0.798	0	0.934	0	0.928	0.938
	slope		-0.0491	0.0001	-0.0001	-0.018	-0.0001	-0.027	0.057	0	0.057	0	5.00E-08	-4.00E-06

Table (1-c): Norfloxacin Derivatives Calculated Descriptors by GAMSS Program

Compound	MIC (µg/ml) or(ppm)	Sum Of Degrees	Sum Of Valence Degrees	Topological Diameter	Total Connectivity	Total Valence Connectivity	Wiener Index	C(4)-C(5)-C(6)	O(14)C carbonyl pyridine charge	F charge	C(8)-C(7)-F(15)	S charge	N(26)-C(25)-N(29)	N(23)-C(22)-S(21)
III	68	64	10 6.6 67	15	3.03 E- 05	3.06E -08	231 5	119.5 25	- 0.56 5	- 0.282	12 0.5 01	0.439 017	125. 397	119.927
IVf	83	82	12 8.6 67	21	1.26 E- 06	6.93E -10	511 7	119.4 37	- 0.55 2	-0.28	12 0.9 9	0.282 965	118. 082	108.501
IVa	84	82	13 2.6 67	21	1.26 E+0 6	3.10E -10	511 7	119.4 96	- 0.55 8	- 0.286	12 0.9 56	0.310 204	117. 874	108.955
IVb	86	84	13 4.6 67	22	8.93 E- 07	2.83E -10	558 9	120.2 84	0.45 - 751 3	- 0.348 074	12 0.3 12	0.245 263	114. 889	110.778
IVd	87	82	12 8.4 44	21	1.26 E- 06	7.86E -10	511 7	119.5 84	- 0.55 447 7	- 0.280 954	12 1.0 91	0.309 275	117. 853	108.933
IVc	88	86	13 4.6 66	22	7.29 E- 07	3.10E -10	606 3	119.4 76	0.56 - 852 1	- 0.284 456	12 1.0 63	0.362 421	117. 895	108.988
IVe	94	82	12 8.6 67	21	1.26 E- 06	6.93E -10	511 7	119.3 43	- 0.55 994	- 0.281 97	11 6.4 66	0.355 84	119. 088	108.656
		dependent variable variation R ²	0. 97 2	0.9 0.8 12	0.00 0.00 001	0.838	0.95 7	0.216	0.01 9	0.012	0.0 41	0.295	0.85 7	0.898
		slope	- 0. 00 5	- 0.0 03	- 3.00 E- 10	4.00E +06	- 3.00 E- 05	0.271	-155	0.2	- 0.2 6	0.379	0.01 1	0.008

Prediction of Most Dependence Properties-Activity

From 38 descriptions there were six highly dependence of properties-activity as shown in figures (1-2) and table (2).

Table (2): Best Regression Properties to Activity of Norfloxacin

Compound	P MIC	η S=-0.069 R ² =0.96 1	B.P (k) S=-0.001 R ² =0.86 7	LogP S=-0.049 R ² =0.833	Refractivit y S=-0.027 R ² =0.884	N Imine-C-N amine S=0.011 R ² =0.857	N Imine-C-S S=0.008 R ² =0.898
III	-1.8325	4.692	870.618	3.45	10.867	125.397	119.927
Ivf	-1.9191	5.779	954.743	5.522	14.561	118.082	108.501
Iva	-1.9243	6.059	977.98	4.645	14.25	117.874	108.955
Ivb	-1.9345	5.83	966.797	4.908	14.714	114.889	110.778
Ivd	-1.9395	6.34	961.839	5.593	14.5882	117.853	108.933
Ivc	-1.9445	6.063	977.625	5.319	15.3931	117.895	108.988
Ive	-1.9731	6.037	973.845	5.863	14.8738	119.088	108.656

η: hardness, S: slope, R2: regression

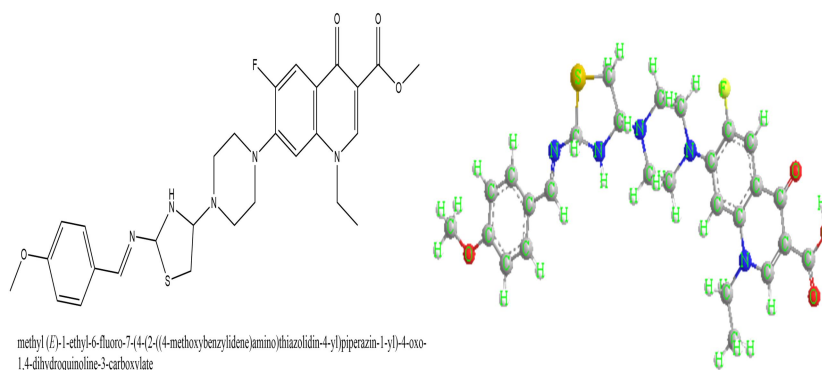


Figure 1: 2D and 3D Molecular Structure of IVb Analog

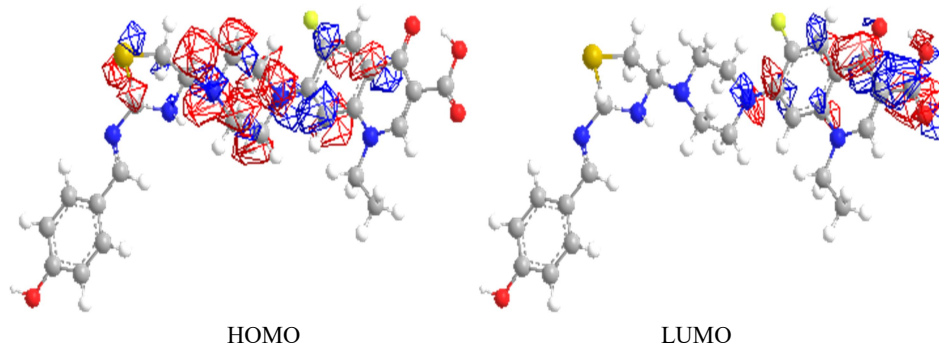


Figure 2: Electronic Structure HOMO and LUMO of Iva Analog

MATHEMATICAL CALCULATION

By using the general equation of QSAR is: [16-17]

$$pMIC = a_0 + \sum a_i X_i$$

Where X_i is p_i * slope with activity

p_i : any property

So the general equation:

$$pMIC = a_0 + a_1 s_1 p_1 + a_2 s_2 p_2 + a_3 s_3 p_3 + a_4 s_4 p_4 + a_5 s_5 p_5 + a_6 s_6 p_6 + \dots$$

For Norfloxacin analogs pMIC were:

$$III- a_0 - 0.323748 a_1 - 0.870618 a_2 - 0.16905 a_3 -$$

$$0.293409 a_4 + 1.379367 a_5 + 1.298902 a_6 = -1.8325$$

$$Ivf- a_0 - 0.398751 a_1 - 0.954743 a_2 - 0.270578 a_3 -$$

$$0.393147 a_4 + 1.298902 a_5 + 0.868008 a_6 = -1.9191$$

$$Iva- a_0 - 0.418071 a_1 - 0.97798 a_2 - 0.228046 a_3 -$$

$$0.38475 a_4 + 1.296614 a_5 + 0.87164 a_6 = -1.9243$$

$$Ivb- a_0 - 0.40227 a_1 - 0.966797 a_2 - 0.240492 a_3 -$$

$$0.3938814 a_4 + 1.263779 a_5 + 0.886224 a_6 = -1.9345$$

$$Ivd- a_0 - 0.43746 a_1 - 0.961839 a_2 - 0.274057 a_3 -$$

$$0.3938814 a_4 + 1.296383 a_5 + 0.871464 a_6 = -1.9395$$

$$Ivc- a_0 - 0.4188347 a_1 - 0.977625 a_2 - 0.260631 a_3 -$$

$$0.4156137 a_4 + 1.296845 a_5 + 0.871904 a_6 = -1.9445$$

$$Ive- a_0 - 0.416553 a_1 - 0.973845 a_2 - 0.287287 a_3 -$$

$$0.4015926 a_4 + 1.309968 a_5 + 0.869248 a_6 = -1.9731$$

The solution of these 7 equations by using Microsoft office excels give:

a_0	0.079512
a_1	-0.08873
a_2	2.347319
a_3	1.334238
a_4	-0.72507
a_5	0.545545
a_6	-0.49027

So theoretical activity for Norfloxacin analogs according to general equation of QSAR;

$$pMIC = 0.079512 + 0.00612237 (\Pi) - 0.002347319 (B.P) - 0.065377662 (\text{LogP}) + 0.75207 (\text{molecular$$

$$\text{refractivity}) + 0.006000995 (\text{N Imine} - \text{C} - \text{Namine angle}) - 0.00392216 (\text{N Imine} - \text{C} - \text{S angle})$$

Now for following compounds theoretical pMIC will be;

$$pMIC (III) = 0.079512 - 0.08873 * -0.323748 + 2.347319 * -$$

$$0.870618 + 1.334238 * -0.16905 - 0.72507 * -$$

$$0.293409 + 0.545545 * 1.379367 - 0.49027 * 1.298902 \quad = -$$

$$1.832500182$$

$$pMIC (Ivf) = 0.079512 - 0.08873 * -0.398751 + 2.347319 * -$$

$$0.954743 + 1.334238 * -0.270578 - 0.72507 * -$$

$$0.393147 + 0.545545 * 1.298902 - 0.49027 * 0.868008 = -$$

$$1.9191003145$$

$$pMIC (Iva) = 0.079512 - 0.08873 * -0.418071 + 2.347319 * -$$

$$0.97798 + 1.334238 * -0.228046 - 0.72507 * -$$

$$0.38475 + 0.545545 * 1.296614 - 0.49027 * 0.87164 \quad = -$$

$$1.924291085$$

$$pMIC (Ivb) = 0.079512 - 0.08873 * -0.40227 + 2.347319 * -$$

$$0.966797 + 1.334238 * -0.240492 - 0.72507 * -$$

$$0.3938814 + 0.545545 * 1.263779 - 0.49027 * 0.886224 \quad = -$$

$$1.934500085$$

$$pMIC (Ivd) = 0.079512 - 0.08873 * -0.43746 + 2.347319 * -$$

$$0.961839 + 1.334238 * -0.274057 - 0.72507 * -$$

$$0.3938814 + 0.545545 * 1.296383 - 0.49027 * 0.871464 \quad = -$$

$$1.939500085$$

$$pMIC (Ivc) = 0.079512 - 0.08873 * -0.4188347 + 2.347319 * -$$

$$0.977625 + 1.334238 * -0.260631 - 0.72507 * -$$

$$0.4156137 + 0.545545 * 1.296845 - 0.49027 * 0.871904 \quad = -$$

$$1.944500085$$

$$pMIC (Ive) = 0.079512 - 0.08873 * -0.416553 + 2.347319 * -$$

$$0.973845 + 1.334238 * -0.287287 - 0.72507 * -$$

$$0.4015926 + 0.545545 * 1.309968 - 0.49027 * 0.869248 \quad = -$$

$$1.973100085$$

DISCUSSION

Calculation on table (1) founds the relation-ship between each property with analog activity and slope of this relation-ship. The positive (+ev) and negative (-ev) proportionality;

two of them geometric properties {N Imine -C-N amine angle (+ev) and N Imine -C-S} angle (+ev), figure (3) and one electronic structure property $\Delta_{\text{LUMO-HOMO}}(\text{I})$ (-ev) figure (2) and three physicochemical properties {B.p., (-ev), log P,

(-ev) and refractivity of molecules, (-ev)} have the best regression (dependent variable variation) selected of R^2 more than 0.83 as shown in Table (3)

Table 3: Practical and Theoretical Norfloxacin Activity

Compound	III	Ivf	Iva	Ivb	Ivd	Ivc	Ive
practical activity	-1.8325	-1.9191	-1.9243	-1.9345	-1.9395	-1.9445	-1.9731
theoretical activity	-1.83250018 2	-1.919100314 5	-1.92429108 5	-1.93450008 5	-1.93950085 1	-1.94450008 5	-1.97310008 5

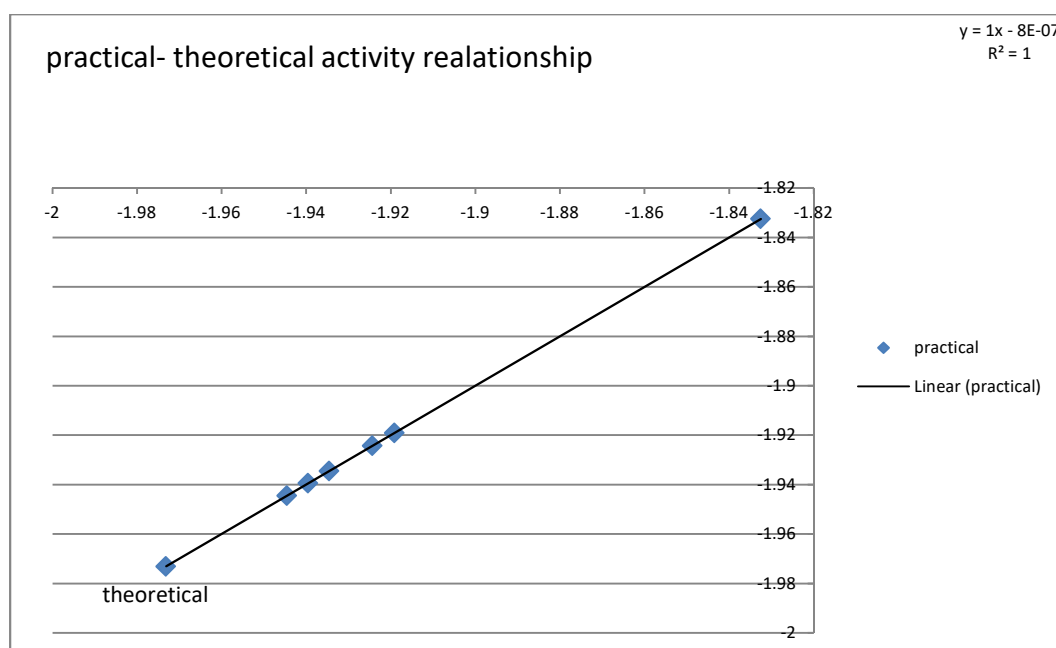


Figure 3: Theoretical and practical Norfloxacin MIC

From table (1) and figure (1) proving that best calculation and predicting of activities and QSAR because the calculated (theoretical) values of activities applicable to practical (experimental) values, So it is excellent method to predicting of novel compound activity.

Conflict of interest: N/A

Informed Consent: the study occurred on animals with the aid of computer software

CONCLUSIONS

QSAR equations in turn are used to predict the activity of new Norfloxacin analog derivatives. Since QSAR produces predictive models derived from application of statistical tools, which correlating with a biological activity included the desirable therapeutic effect with descriptors of molecular structure or properties. In general, the good quality of QSAR equation depends on many factors such as; quality of input data, selection of descriptors and a statistical methods for validation of model. Ultimately, QSAR would lead to robust equation and predictive models are capable of making accurate and reliable calculations for a new Norfloxacin analog.

Ethical approval: The research involves a computerized study and not involved any in vitro or in vivo experiments, so; not required approval from the ethical committee.

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