

# Quantitative Structure Property Relationship Analysis of Kulli's Topological Indices

# **D.G.Prakasha<sup>1</sup>**, and Manjunatha Gali<sup>2</sup>

<sup>1</sup>Department of Mathematics, Davangere University, ShivagangotriCampus, Davangere-577 007, India. E-mail:prakashadg@gmail.com, prakashadg@davangereuniversity.ac.in

<sup>2</sup>Department of Mathematics, Shri Gavisiddeshwara Arts, Science and Commerce College, Gavimath Campus, Koppal – 583231, India. E-mail:<u>manjugalijack@gmail.com</u>

#### Abstract:

Molecular descriptor is a number associated with the moleculargraph. The applications of molecular descriptors in predicting the physico-chemico properties of chemical compounds framed a remarkable benchmark in the field of mathematical chemistry. In this paper, we consider Kulli's molecular descriptors to examine their applications in QSPR(Quantitative Structure Property Relationship) study of alkanes. The results of QSPR-analysis demonstrated F-Somborindex (FSLI), sum connectivity Gourava index (SCGI), product connectivity Gourava index (PCGI), multiplicative atom bond sum connectivity index (MASCI) and Revan Sombor index (RSI) shows good correlation with the physical properties of alkanes.

**Keywords:** F-Sombor index, multiplicative atom bond sum connectivity index, QSPR *AMS* **Subject Classification:** 05C90; 05C12.

#### 1. Introduction

A graph is said to be simple if it doesn't contain multiple edges or loops. Throughoutthispapersimpleandundirectedgraphsareconsidered with vertex set *V* and edges et *E*. T he order and size of *G* is denoted by |V| = n and |E| = m respectively. The degree of avertex  $v \in V$  is the number of edges incident to v and it is denoted by  $d_G(v)$ . The degree of an edge e = uv is defined as  $d_G(e) = d_G(u) + d_G(v) - 2$ . For undefined terminology in this paper refer[3].

Topological index is simply a numerical associated with the molecular graph. So far, large number of such quantities are put forward by many researchers right from 1972.

One of the most useful topological indices are the Zagreb indices [2] which are defined as:

$$M_1(G) = \sum_{i=1}^n d_G(v)^2 ,$$
$$M_2(G) = \sum_{u,v \in E(G)} d_G(u) d_G(v)$$

,

Where  $M_1$  and  $M_2$  are the first and second Zagreb indices respectively.

Motivated by degree based topological indices, Kulli[4- 0] put forward the following topological indices:

SI.	Molecular descriptor	Abbreviation	Definition
No.			
1	F-Sombor Index [6]	FSLI	$\sum_{uv \in E(G)} \sqrt{(d(u)^4 + d(v)^4)}$
2	Sum Connectivity Gourava Index [9]	SCGI	$\sum_{uv \in E(G)} \frac{1}{\sqrt{(d(u) + d(v)) + (d(u) * d(v))}}$
3	Product Connectivity Gourava Index [11]	PCGI	$\sum_{uv \in E(G)} \frac{1}{\sqrt{(d(u) + d(v)) * (d(u) * d(v))}}$
4	Multiplicative Atom Bond Sum Connectivity Index [8]	MASCI	$\sum_{uv \in E(G)} \sqrt{\frac{d(u) + d(v) - 2}{d(u) + d(v)}}$
5	Sum Augmented Index [10]	SAI	$\sum_{uv \in E(G)} \left( \frac{d(u) + d(v)}{d(u) + d(v) - 2} \right)^3$
6	Revan Sombor Index [4]	RSI	$\sum_{uv \in E(G)} \sqrt{\left(\Delta(G) + \delta(G) - d(u)\right)^2 + \left(\Delta(G) + \delta(G) - d(v)\right)^2}$

#### Table 1: Molecular Descriptors

### 2. Applications of Kulli's Molecular Descriptors

Here we consider the topological indices depicted in Table1 along with eight representative physical properties [boiling points(BP), molar volumes(mv) at 20°C, molar refractions(mr) at 20°C, heats of vaporization(hv) at 25°C, surface tensions(st) 20°C and melting points (mp)] of the 67 alkanes from *n*-butanes to nonanes. Values for this property were taken from[11]. The above said distance-based topological indices and the experimental values for the physical properties of 67 alkanes are listed in Table2.

#### SI. Alkane ct cp(at st(dyne bp hv mv mr mp No /cm) $({}^{0}C)$ $(cm^3)$ (kJ) $({}^{0}C)$ m) $({}^{0}C)$ $(cm^3)$ 1 Butane -0.500 152.01 37.47 -138.35 2 -11.730 134.98 -159.60 2-methyl propane 36 3 Pentane 36.074 115.205 25.2656 196.62 33.31 -129.72 26.42 16.00 4 2-methyl butane 25.2923 32.9 15.00 27.852 116.426 24.59 187.70 -159.90 5 2,2 dimethylpropane 9.503 112.074 25.7243 160.60 31.57 -16.55 21.78 Hexane 6 130.688 234.70 68.740 29.9066 31.55 29.92 18.42 -95.35 7 29.95 2-methylhexane 60.271 131.933 29.9459 29.86 224.90 17.38 -153.67 8 3-methylhexane 63.282 129.717 29.8016 30.27 231.20 30.83 18.12 -118.28 2,2-methylbutane 49.741 16.30 -99.87 9 132.744 29.9347 27.69 216.20 30.67 10 2,3-dimethylbutane 57.988 130.240 29.8104 29.12 227.10 30.99 17.37 -128.54 27.01 11 Heptane 98.427 146.540 34.5504 36,55 267.55 20.26 -90.61 12 2-methylhexane 90.052 147.656 34.5908 34.80 257.90 27.2 19.29 -118.28 13 3-methylhexane 91.850 145.821 34.4597 35.08 262.40 28.1 19.79 -119.40 14 93.475 35.22 28.6 20.44 3-ethylpentane 143.517 34.2827 267.60 -118.60 15 2,2-dimethylpentane 79.197 148.695 34.6166 32.43 247,70 28.4 18.02 -123.81 16 2,3-dimethylpentane 89.784 144.153 34.3237 34.24 264.60 29.2 19.96 -119.10 17 2,4-dimethylpentane 80.500 148.949 34.6192 32.88 247.10 27.4 18.15 -119.24 18 3,3-dimethylpentane 86.064 144.430 34.3323 33.02 263.00 30 19.59 -134.46 19 Octane 125.665 296.20 -56.79 162.592 39.1922 41.48 24,64 21.76 20 2-methylheptane 117.647 163.663 39.2316 288.00 24.8 20.60 -109.04 39.68 21 3-methylheptane 282.00 25.6 -120.50 118.925 161.832 39.1001 39.83 21.17 22 4-methylheptane 117.709 162.105 39.1174 39.67 290.00 25.6 21.00 -120.95 23 3- ethylhexane 118.53 160.07 38.94 39.40 292.00 25.74 21.51 24 37.29 2,2-dimethylhexane 10.84 164.28 37.29 279.00 25.6 19.60 -121.18 25 2,3-dimethylhexane 115.607 38.79 20.99 160.39 38.79 293.00 26.6 2,4-dimethylhexane 109.42 37.76 26 163.09 37.76 282.00 25.8 20.05 -137.50 2,5-dimethylhexane 109.10 164.69 37.86 279.00 19.73 -91.20 27 37.86 25 28 160.87 37.93 37.93 290.84 27.2 -126.10 3,3- dimethylhexane 111.96 20.63 21.64 29 3,4-dimethylhexane 117.72 158.81 39.02 39.02 298.00 27.4 30 3-ethyl-2-115.65 158.79 38.83 38.52 295.00 27.4 21.52 -114.96 methylpentane 3-ethyl-3-157.02 38.71 37.99 28.9 21.99 -90.87 31 118.25 305.00 methylpentane 32 2.2.3-109.84 159.52 38.92 36.91 294.00 28.2 20.67 -112.27 trimethylpentane 33 2,2,4-99.23 39.26 271.15 25.5 -107.38 165.08 35.13 18.77 trimethylpentane 34 2,3,3-157.29 38.76 37.22 303.00 29 -100.70 114.76 21.56 trimethylpentane

# **Table 2. Physical Properties of alkanes**

07	2.2.4	110.14	150.05	20.05	07.61	205.00		01.1.1	100.01
35	2,3,4-	113.46	158.85	38.86	37.61	295.00	27.6	21.14	-109.21
	trimethylpentane		1 - 0 - 1	10.01					
36	Nonane	150.79	178.71	43.84	46.44	322.00	22.74	22.92	-53.52
37	2-methyloctane	143.26	179.77	43.87	44.65	315.00	23.6	21.88	-80.40
38	3-methyloctane	144.18	177.95	43.72	44.75	318.00	23.7	22.34	-107.64
39	4-methyloctane	142.48	178.15	43.76	44.75	318.30	23.06	22.34	-113.20
40	3-ethylheptane	143.00	176.41	43.64	44.81	318.00	23.98	22.81	-114.90
41	4-ethylheptane	141.20	175.68	43.49	44.81	318.30	23.98	22.81	
42	2,2-dimethylheptane	132.69	180.50	43.91	42.28	302.00	22.8	20.80	-113.00
43	2,3-dimethylheptane	140.50	176.65	43.63	43.79	315.00	23.79	22.34	-116.00
44	2,4-dimethylheptane	133.50	179.12	43.73	42.87	306.00	22.7	23.30	
45	2,5-dimethylheptane	136.00	179.37	43.84	43.87	307.80	22.7	21.30	
46	2,6-dimethylheptane	135.21	180.91	43.92	42.82	306.00	23.7	20.83	-102.90
47	3,3-dimethylheptane	137.300	176.897	43.6870	42.66	314.00	24.19	22.01	
48	3,4-dimethylheptane	140.600	175.349	43.5473	43.84	322.70	24.77	22.80	
49	3,5-dimethylheptane	136.000	177.386	43.6379	42.98	312.30	23.59	21.77	
50	4,4-dimethylheptane	135.200	176.897	43.6022	42.66	317.80	24.18	22.01	
51	3-ethyl-2-	138.000	175.445	43.6550	43.84	322.70	24.77	22.80	
	methylhexane								
52	4-ethyl-2-	133.800	177.386	43.6472	42.98	330.30	25.56	21.77	
	methylhexane								
53	3-ethyl-3-	140.600	173.077	43.2680	44.04	327.20	25.56	23.22	
	methylhexane								
54	2,2,4-	126.540	179.220	43.7638	40.57	301.00	23.39	20.51	-120.00
	trimethylhexane								
55	2,2,5-	124.084	181.346	43.9356	40.17	296.60	22.41	20.04	-105.78
	trimethylhexane								
56	2,3,3-	137.680	173.780	43.4347	42.23	326.10	25.56	22.41	-116.80
	trimethylhexane								
57	2,3,4-	139.00	173.498	43.4917	42.93	324.20	25.46	22.80	
	trimethylhexane								
58	2,3,5-	131.340	177.656	43.6474	41.42	309.40	23.49	21.27	-127.80
	trimethylhexane								
59	3,3,4-	140.460	172.055	43.3407	42.28	330.60	26.45	23.27	-101.20
	trimethylhexane								
60	3,3-diethylpentane	146.168	170.185	43.1134	43.36	342.80	26.94	23.75	-33.11
61	2,2-dimethyl-3-	133.830	174.537	43.4571	42.02	322.60	25.96	22.38	-99.20
	ethylpentane				-		-		-
62	2,3-dimethyl-3-	142.000	170.093	42.9542	42.55	338.60	26.94	23.87	
	ethylpentane								
63	2,4-dimethyl-3-	136.730	173.804	43.4037	42.93	324.20	25.46	22.80	-122.20
	ethylpentane								
64	2,2,3,3-	140.274	169.495	43.2147	41.00	334.50	27.04	23.38	-99.0
	tetramethylpentane			-					-
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65	2,2,3,4-	133.016	173.557	43.4359	41.00	319.60	25.66	21.98	-121.09
	tetramethylpentane								
66	2,2,4,4-	122.284	178.256	43.8747	38.10	301.60	24.58	20.37	-66.54
	tetramethylpentane								
67	2,3,3,4-	141.551	169.928	43.2016	41.75	334.50	26.85	23.31	-102.12
	tetramethylpentane								

### 3. Regression Models

The following statistical models have been used for the study:

- Linear Model: P = aX + b,
- Quadratic Model:  $P = aX^2 + bX + c$ ,
- Logarithmic Model:  $P = a + b \ln(x)$ ,

where P is a physical property, X is a domination parameter, a, b and c are constants. Now, we have obtained the following different regression models for each parameter, which are listed below.

#### 4. F-Sombor Index(FSLI)

- 1. Linear Model
- bp = 2.055 + [FSLI]29.5 (1)
- mv = -1.104 + [FSLI]33.5 (2)
- mr = 0.187 + [FSLI]26.3 (3)
- $\Box v = 0.189 + [FSLI]25.5$ (4)
- ct = 3.469 + [FSLI]177.9 (5)
- cp = 39.875 [FSLI]6.4 (6)
- st = 0.098 + [FSLI]18.7 (7)
- mp = -123.759 + [FSLI]0.457 (8)
- 2. Quadratic Model

$$bp = 5.9[FSLI]^2 - 0.02[FSLI] - 50.6$$
(9)

$$mv = 2.4[FSLI]^2 - 0.007 [FSLI] + 91.1$$
(10)

$$mr = 0.3[FSLI]^2 - 0.003[FSLI] + 19.2$$
(11)

$$hv = 0.4[FSLI]^2 - 0.004[FSLI] + 20.1$$
(12)

 $ct = 4.9[FSLI]^2 - 0.09[FSLI] + 105.7$ (13)

$$cp = -0.4[FSLI]^2 + 0.003[FSLI] + 38.2$$
(14)

$$st = 0.3[FSLI]^2 - 0.002[FSLI] + 15.1$$
(15)

$$mp = 0.7[FSLI]^2 - 0.007[FSLI] + 149.2$$
(16)

3. Logarithmic Model

bp = 64.9 - ln[FSLI] 167.4 (1)	7	)
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mv = 66.8 + ln[FSLI]74.5	(18)
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$$mr = 8.9 + ln[FSLI]27.1$$
 (19)

hv = 11.4 + ln[FSLI]23.2 (20)

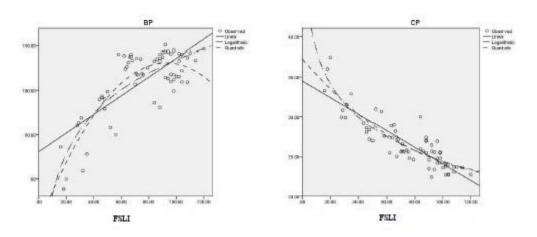
$$ct = 12.07 - ln[FSLI]9.1$$
 (21)

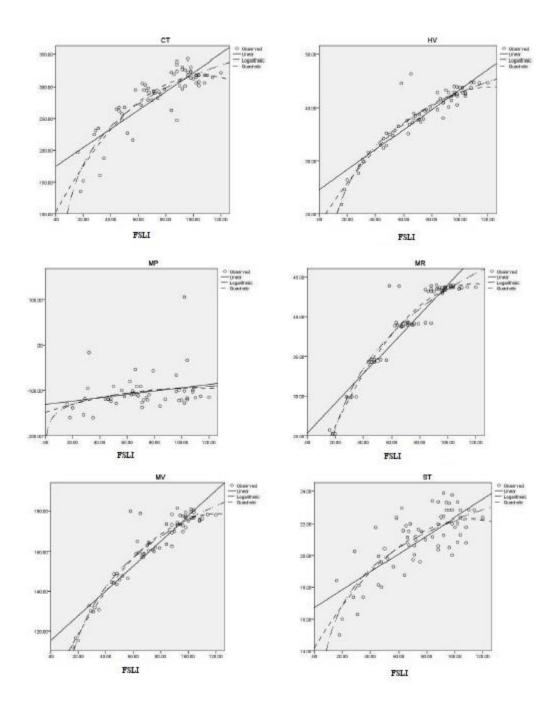
$$cp = -8.06 - ln[FSLI]51.3$$
 (22)

$$st = 10.06 + ln[FSLI]8.3$$
 (23)

$$mp = 21.1 - ln[FSLI]199.7$$
(24)

# In the following figures, the correlation of F-Sombor Index with above mentioned physical properties of alkanes are shown:





### 5. Sum Connectivity Gourava Index (SCGI)

1. Linear Model

$$bp = -46.450 + [SCGI]0.379 \tag{25}$$

$$mv = 78.533 + [SCGI]0.283 \tag{26}$$

- $mr = 10.916 + [SCGI]1.412 \tag{27}$
- $hv = 14.435 + [SCGI]10.457 \tag{28}$
- ct = 99.561 + [SCGI]10.457 (29)
- cp = 28.113 [SCGI]0.235 (30)
- st = 25.256 + [SCGI]0.529 (31)
- $mp = -174.768 + [SCGI]2.436 \tag{32}$
- 2. Quadratic Model

$$bp = 10.567[SCGI]^2 - 0.884[SCGI] - 132.568$$
(33)

$$mv = 7.876[SCGI]^2 - 0.143[SCGI] + 60.263$$
(34)

$$mr = 3.132[SCGI]^2 - 0.052[SCGI] + 31.184$$
(35)

$$hv = 3.448[SCGI]^2 - 0.075[SCGI] + 11.971$$
(36)

 $ct = 36.365[SCGI]^2 - 0.625[SCGI] + 18.542$ (37)

$$cp = -7.363[SCGI]^2 + 0.047[SCGI] + 56.692$$
(38)

$$st = 0.967[SCGI]^2 - 0.010[SCGI] + 13.482$$
 (39)

$$mp = 2.732[SCGI]^2 - 0.047[SCGI] - 149.002$$
(40)

### 3. Logarithmic Model

$$bp = -194.741 - ln[SCGI]100.176 \tag{41}$$

$$mv = 9.017 + \ln[SCGI]64.949 \tag{42}$$

$$mr = -3.628 + ln[SCGI]19.187 \tag{43}$$

$$hv = -7.376 + \ln[SCGI]12.107 \tag{44}$$

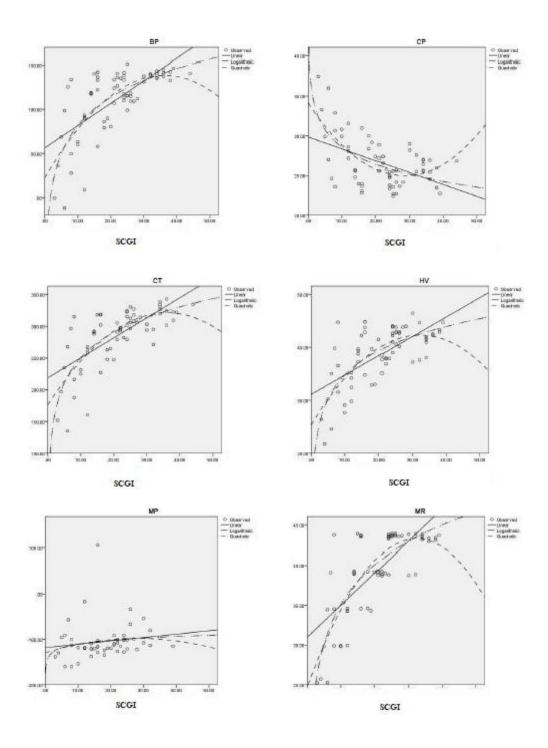
$$ct = -117.478 + ln[SCGI]165.712 \tag{45}$$

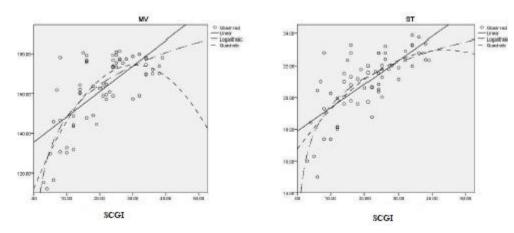
$$cp = 55.929 - ln[SCGI]10.959 \tag{46}$$

$$st = 6.786 + ln[SCGI]7.458$$
 (47)

$$mp = -210.472 + ln[SCGI]35.081 \tag{48}$$

In the following figures, the correlation of sum connectivity Gourava index with abovementioned physical properties of alkanes are shown:





# 6. Product Connectivity Gourava Index (PCGI)

- 1. Linear Model
  - $bp = 5.194 + [PCGI]4.554 \tag{49}$
  - $mv = 107.456 + [PCGI]0.870 \quad (50)$
  - mr = 23.218 + [PCGI]1.456 (51)
  - $hv = 23.909 + [PCGI]0.886 \tag{52}$
  - ct = 123.038 + [PCGI]4.317 (53)
  - cp = 57.082 [PCGI]0.248 (54)
  - st = 13.125 + [PCGI]0.578 (55)
  - mp = -145.961 + [PCGI]1.265(56)
- 2. Quadratic Model

$$bp = 11.9[PCGI]^2 - 0.12[PCGI] - 47.612$$
 (57)

$$mv = 4.654[PCGI]^2 - 0.042[PCGI] + 88.216$$
 (58)

 $mr = 1.132[PCGI]^2 - 0.15[PCGI] + 18.132$  (59)

$$hv = 1.316[PCGI]^2 - 0.015[PCGI] + 14.259$$
(60)

$$ct = 10.192[PCGI]^2 - 0.0101[PCGI] + 78.430$$
 (61)

$$cp = -0.980[PCGI]^2 + 0.014[PCGI] + 23.965 \quad (62)$$

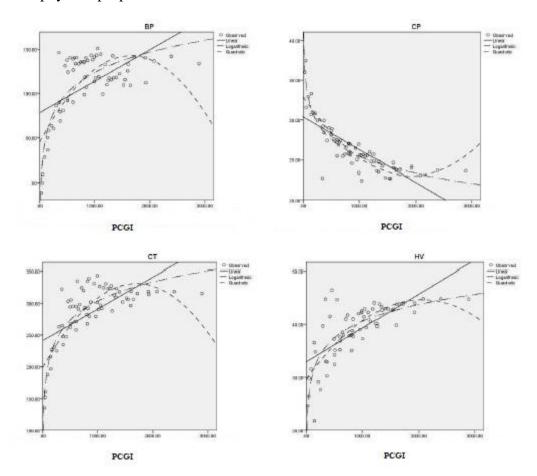
$$st = 0.340[PCGI]^2 - 0.008[PCGI] + 23.456$$
(63)

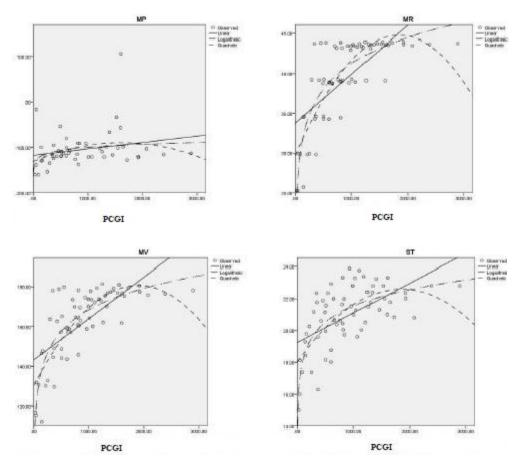
$$mp = 2.613[PCGI]^2 - 0.25[PCGI] - 134.608$$
(64)

3. Logarithmic Model

bp = -147.275 - ln[PCGI]78.903	(65)
mv = 45.434 + ln[PCGI]40.210	(66)
mr = 0.876 + ln[PCGI]10.236	(67)
hv = 21.608 + ln[PCGI]0.655	(68)
ct = -50.591 + ln[PCGI]100.567	(69)
cp = 44.853 - ln[PCGI]4.139	(70)
st = 5.452 + ln[PCGI]3.511	(71)
mp = -186.817 + ln[PCGI]18.221	(72)

In the following figures, the correlation of product connectivity Gourava index with above mentioned physical properties of alkanes are shown:





# 7. Multiplicative Atom Bond Sum Connectivity Index(MASCI)

1. Linear Model

$$bp = -56.617 + [MASCI]36.785$$
(73)

$$mv = 75.043 + [MASCI]13.988 \tag{74}$$

$$mr = 14.070 + [MASCI]5.401 \tag{75}$$

$$hv = 16.809 + [MASCI]5.912 \tag{76}$$

- ct = 74.987 + [MASCI]35.537 (77)
- cp = 37.578 [MASCI]3.013 (78)
- st = 15.540 + [MASCI]2.435 (79)

$$mp = -158.005 + [MASCI]10.104$$
(80)

2. Quadratic Model

$$bp = 78.544[MASCI]^2 - 4.622[MASCI] - 168.756$$
(81)

$$mv = 51.766[MASCI]^2 - 4.004[MASCI] + 14.767$$
(82)

$$mr = 10.465[MASCI]^2 - 0.178[MASCI] - 1.474$$
(83)

$$hv = 12.506[MASCI]^2 - 1.050[MASCI] + 4.654$$
(84)

$$ct = 124.251[MASCI]^2 - 8.838[MASCI] - 110.078$$
 (85)

$$cp = -10.902[MASCI]^2 + 0.879[MASCI] + 56.177$$
(86)

$$st = 3.676[MASCI]^2 - 0.253[MASCI] + 5.254$$
 (87)

$$mp = 12.231[MASCI]^2 - 0.881[MASCI] - 156.446$$
 (88)

3. Logarithmic Model

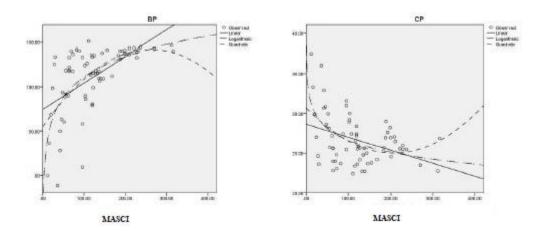
$$bp = -165.05 - ln[MASCI] 132.211$$
(89)

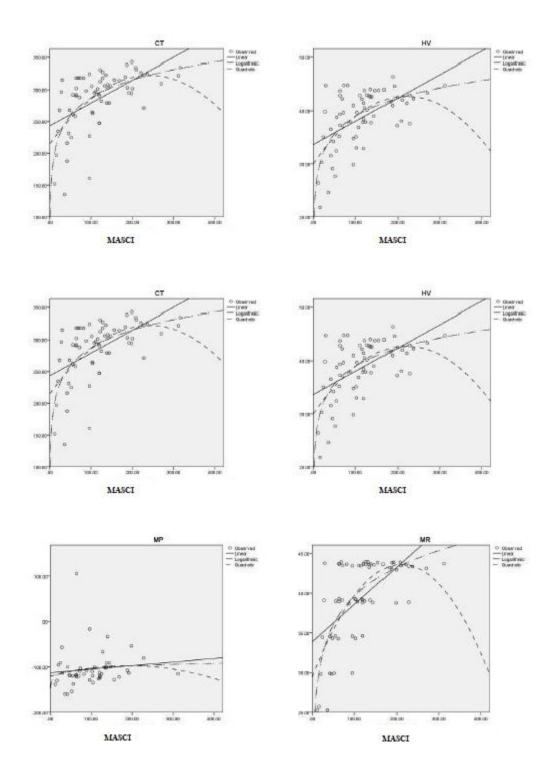
- mv = 45.460 + ln[MASCI]56.637 (90)
- mr = 0.876 + ln[MASCI] 12.235 (91)
- hv = 7.204 + ln[MASCI]16.929(92)
- ct = -55.591 + ln[MASCI]110.013 (93)
- cp = 42.629 ln[MASCI]10.323 (94)

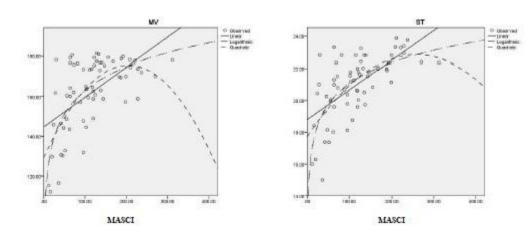
$$st = 10.567 + ln[MASCI]4.725$$
 (95)

$$mp = -162.580 + ln[MASCI]47.054$$
(96)

In the following figures, the correlation of multiplicative atom bond sum connectivity index with above mentioned physical properties of alkanes are shown:







- 8. Sum Augmented Index (SAI)
  - 1. Linear Model

bp = b	11.536 +	[ <i>SAI</i> ]1.351	(97)
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- mv = 110.932 + [SAI]1.086 (98)
- mr = 30.569 + [SAI]0.213 (99)

 $hv = 21.315 + [SAI]0.328 \tag{100}$ 

- $ct = 151.329 + [SAI]3.201 \tag{101}$
- cp = 32.467 [SAI]0.280 (102)

$$st = 14.205 + [SAI]0.199$$
 (103)

 $mp = -131.576 + [SAI]0.276 \tag{104}$ 

2. Quadratic Model

$$bp = 5.23[SAI]^2 - 0.036[SAI] - 78.089$$
(105)

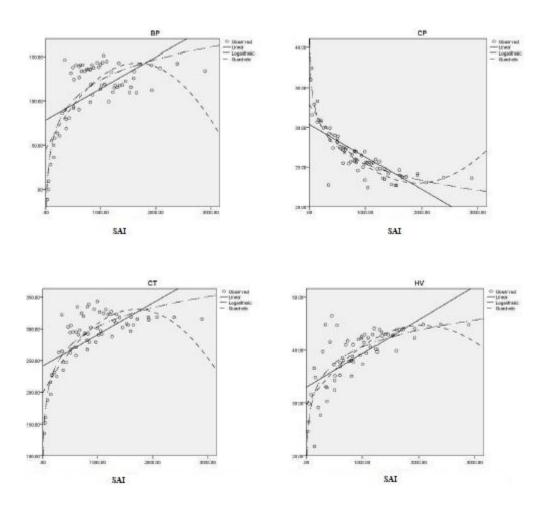
- $mv = 1.446[SAI]^2 0.008[SAI] + 101.796 \quad (106)$
- $mr = 0.625[SAI]^2 0.001[SAI] + 19.575$ (107)
- $hv = 0.615[SAI]^2 0.001[SAI] + 22.595$ (108)
- $ct = 6.22[SAI]^2 0.035[SAI] + 55.987$ (109)
- $cp = -0.979[SAI]^2 + 0.002[SAI] + 28.205$ (110)
- $st = 0.416[SAI]^2 0.001[SAI] + 13.182$ (111)

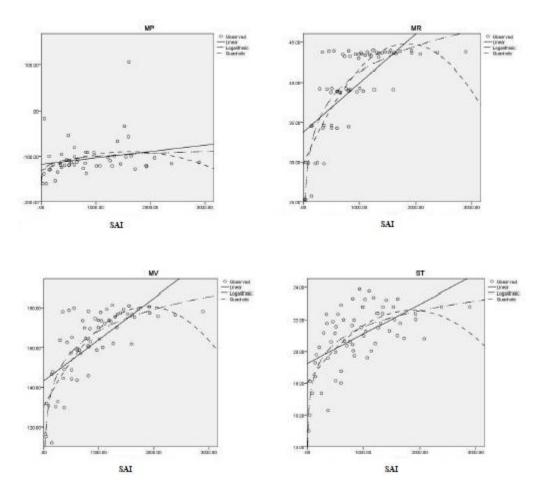
$$mp = -2.241[SAI]^2 - 0.053[SAI] - 74.870 \quad (112)$$

3. Logarithmic Model

bp = -177.566 + ln[SAI]80.881	(113)
mv = 27.459 + ln[SAI]34.302	(114)
mr = -0.988 + ln[SAI]10.396	(115)
hv = 0.359 + ln[SAI]10.402	(116)
ct = -131.812 + ln[SAI]106.234	(117)
cp = 49.847 - ln[SAI]49.847	(118)
st = 9.067 + ln[SAI]3.241	(119)
mp = -148.213 + ln[SAI]16.200	(120)

In the following figures, the correlation of sum augmented index with above mentioned physical properties of alkanes are shown:





- 9. Revan Sombor Index (RSI)
  - 1. Linear Model

$$bp = -43.940 + [RSI]24.445 \tag{121}$$

 $mv = 78.145 + [RSI]10.557 \tag{122}$ 

$$mr = 17.101 + [RSI]4.777 \tag{123}$$

$$hv = 14.673 + [RSI]2.454 \tag{124}$$

$$ct = 70.271 + [RSI]31.564$$
 (125)

- cp = 39.667 [RSI]2.102 (126)
- st = 12.563 + [RSI]2.01 (127)

$$mp = -149.282 + [RSI]6.457 \tag{128}$$

2. Quadratic Model

$$bp = 82.320[RSI]^2 - 3.141[RSI] - 200.458$$
(129)

$$mv = 10.019[RSI]^2 + 0.123[RSI] + 78.479$$
(130)

$$mr = 3.624[RSI]^2 - 0.013[RSI] + 13.501$$
(131)

$$hv = 2.234[RSI]^2 + 0.061[RSI] + 17.346$$
(132)

$$ct = 61.467[RSI]^2 - 2.567[RSI] - 10.167$$
(133)

$$cp = -3.002[RSI]^2 + 0.078[RSI] + 43.276$$
(134)

$$st = 2.754[RSI]^2 - 0.134[RSI] + 10.012$$
 (135)

$$mp = -14.348[RSI]^2 + 1.998[RSI] - 87.176$$
(136)

3. Logarithmic Model

$$bp = -139.117 + ln[RSI] 136.276 \tag{137}$$

$$mv = 42.457 + ln[RSI]64.795$$
(138)

$$mr = 3.356 + ln[RSI]$$
18.456 (139)

$$hv = 4.367 + ln[RSI]17.187 \tag{140}$$

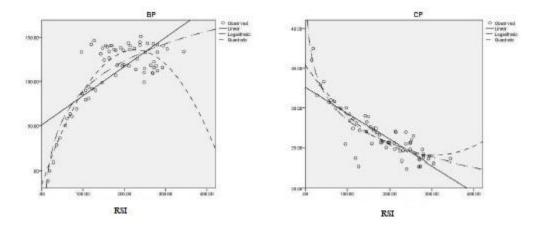
$$ct = -39.742 + ln[RSI]177.812$$
(141)

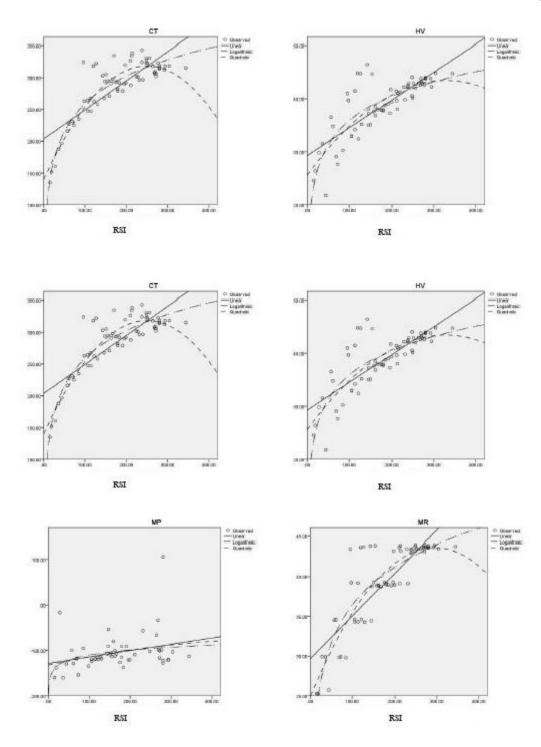
$$cp = 46.467 - ln[RSI]10.419 \tag{142}$$

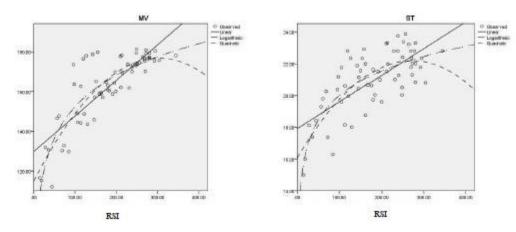
$$st = 8.999 + ln[RSI]7.672$$
 (143)

$$mp = -189.470 + ln[RSI]37.627$$
(144)

In the following figures, the correlation of Revan Sombor index with above mentioned physical properties of alkanes are shown:







**10. Discussion and Concluding Remark** 

By inspection of the data given in tables 1 and 2, and equations (7)-(150) it is possible to draw a number of conclusions for the given domination parameters.

- The F-Sombor index(FSLI) is the first has correlation coefficient value ranging from 0.071 to 0.88 for linear model with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for molar volume of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.084 to 0.9 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for molar refraction of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.074 to 0.921 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for molar refraction of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for molar refraction of alkanes respectively.
- The sum connectivity Gourava index(SCGI) plays an important role in QSPR studies of alkanes. The correlation coefficient value ranges from 0.023 to 0.523 for linear model with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for molar refractions of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.091 to 0.76 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for molar refractions of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.054 to 0.78 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes respectively.

- The product connectivity Gourava index (PCGI) shows better predicting power than the connected domination number. The correlation coefficient value ranges from 0.055 to 0.698 for linear model with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical pressure of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.077 to 0.891 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical pressure of alkanes and maximum correlation coefficient value for critical pressure of alkanes and maximum correlation coefficient value for critical pressure of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.088 to 0.895 with minimum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for critical pressure of alkanes respectively.
- The multiplicative atom bond sum connectivity Index (MASCI) found to be useful parameter in QSPR-studies. The correlation coefficient value ranges from 0.092 to 0.867 for linear model with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical pressure of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.098 to 0.897 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical temperature of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.09 to 0.923 with minimum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes respectively.
- The sum augmented index (SAI) found to be less useful domination parameter compare to the total domination number. The correlation coefficient value ranges from 0.035 to 0.478 for linear model with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for surface tension of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.033 to 0.779 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical temperature of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.021 to 0.793 with minimum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for critical temperature of alkanes respectively.
- The Revan Sombor index (RSI) has a good predictive power for the set of alkanes. The correlation coefficient value ranges from 0.086 to 0.967 for linear model with minimum

correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for critical temperature of alkanes respectively. For quadratic model the correlation coefficient value ranges from 0.091 to 0.578 with minimum correlation coefficients value for melting points of alkanes and maximum correlation coefficient value for molar refractions of alkanes respectively. Finally, for logarithmic model the correlation coefficient value ranges from 0.062 to 0.578 with minimum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes and maximum correlation coefficient value for melting points of alkanes respectively.

#### 11. Conclusion.

From practical point of view, topological indices for which the absolute value of correlation coefficient is less than 0.8 can be characterized as useless. Overall, the molecular descriptors considered for this study are useful candidates to predict the critical temperature of alkanes.

# References

[1] BiyeRen, Atom-Type-Based AI Topological Descriptors: Application in Structure-Boiling Point Correlation of Oxo Organic Compound, *J. Chem. Inf. Comput. Sci.*,43 (2003), 1121-1131.

[2]I. Gutman, N. Trinajsti'c, Graph theory and molecular orbitals. Total  $\pi$ -electron energy of alternant hydrocarbons, Chem. Phys. Lett., 17 (1972), 535–538.

[3] F. Harary (1969), Graph Theory, Addison–Wesely, Reading.

[4] V. R. Kulli, G. N. Adithya, N. D. Soner, Gourava Indices of Certain Windmill Graphs, *International Journal of Mathematics Trends and Technology*, Vol. 68 Issue 9, (2022) 51-59.

[5] V. R. Kulli and Ivan Gutman, Revan Sombor Index, *Journal of Mathematics and Informatics*, Vol. 22, (2022) 23-27.

[6] V. R. Kulli, F-Sombor and Modified F-Sombor Indices of certain Nanotubes, vol. 27, No. 1, (2023) 13-17.

[7] V. R. Kulli, Gourava Nirmala Indices of Certain Nanostructures, *International Journal of Mathematical Archive*-14(2), 1-9, (2023) 2229-5046.

[8] V. R. Kulli Multiplicative Atom Bond Sum Connectivity Index of Certain Nanotubes, Vol 27,No. 1, (2023) 31-35.

[9] V.R.Kulli, On the Sum Connectivity GouravaIndex, *International Journal of Mathematical Archive*, vol. 8, no. 6, (2017) 211-217.

[10] V.R.Kulli, The Product Connectivity Gourava Index, *Journal of Computer and Mathematical Sciences*, vol. 8, no. 6, (2017) 235-242.

[11] Shailaja Shirakol, Manjula Kalyanshetti, Sunilkumar M. Hosamani, QSPR Analysis of certain Distance Based Topssological Indices, *Applied Mathematics and Nonlinear Sciences* 4(2), 371-386(2019).