

ANALYTICAL METHOD DEVELOPMENT AND VALIDATION FOR THE DETERMINATION OF METHYLCOBALAMIN BY UFLC IN ORAL SOLID DOSAGE FORMS

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Abstract:

Methylcobalamin Tablets are available in the market. But the literature review reveals that there is no official monograph for Methylcobalamin Tablets in Indian Pharmacopoeia and also no analytical method reported for the analysis of Methylcobalamin by UFLC in oral solid dosage form. Spectrophotometer and RP-HPLC are the reported analytical methods for compounds either individually or in combination with other dosage form. Hence, it was felt that, there is a need of new analytical method development for the estimation of Methylcobalamin by UFLC in oral solid dosageform. Present work is aimed to develop a new, simple, fast, rapid, accurate, efficient and reproducible Liquid Chromatography method for the analysis Methylcobalamin by UFLC in solid dosage form. The developed method will be validated according to ICH guidelines. Retention time of Methylcobalamin was found to be 1.193min. and the regression equation of Methylcobalamin is y=0.0000x-0.1035 &r²=0.9999. The Method precision is 0.73%, Intermediate precision for day 01 is 052% and the intermediate precision for Day 02 is 0.55% The assay of Methylcobalamin found as 100.21%.Since the Retention time of Methylcobalamin is less than 2.0 minutes, the developed method is cost effective with short analytical time. So the method developed was simple and economical that can be adopted in regular Quality control test inIndustries.

Key words: Methylcobalamin, Oral solid dosage forms ,RP-HPLC, UFLC

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Introduction

Methylcobalamin Tablets are available in the market. But the literature review reveals that there is no official monograph for Methylcobalamin Tablets in Indian Pharmacopoeia and also no analytical method reported for the analysis of Methylcobalaminby UFLC in oral solid dosage form.

Spectrophotometer and RP-HPLC are the reported analytical methods for compounds either individually or in combination with other dosage form. Hence, it was felt that, there is a need of new analytical method development for the estimation of Methylcobalaminby UFLC in oral solid dosageform.

Present work is aimed to develop a new, simple, fast, rapid, accurate, efficient and reproducible Liquid Chromatography method for the analysis Methylcobalaminby UFLC in solid dosage form. The developed method will be validated according to ICH guidelines.

Materials and Methods:

Materials: Methylcobalamin tablet prepared in house, HPLCwater, ortho phosphoric acid, Acetonitrile, Ammonia. All the above chemicals and solvents are from Rankem.

Instruments:

- ElectronicsBalance-Mettler toledo
- pHmeter Hanna, 5 Pointcalibration
- Ultrasonicator-BVKenterprises
- LC-20AD PROMINENCEUFLC
- Software: LC SOLUTIONSHIMADZU

Methods:

Chromatographic condition:

- 1. Column: INERTSIL ODS, C18 (50 mm X 3.6 mm,3µm)
- 2. Wavelength:352nm
- 3. Flowrate:0.7ml/min
- 4. Injection volume:2µl
- 5. Mobile phase: Mixture of Buffer: Acetonitrile(600:400)

Buffer preparation: 3.4ml of orthophosphoric acid in 1000 ml of water PH to 2.7 with ammonia. **Diluent**: Based up on the solubility of the drugs, diluent was selected, Mobile phaseused as diluent. **Preparation of Standard solution**: Weigh accurately and transfer about 10 mg of Methylcobalaminstandard into 100 ml volumetric standard flask. Dissolveand makeup to the volume with Mobilephase.

Preparation of Sample solution: Weigh accurately equivalent to 10mg of sample into 100

ml volumetric standard flask. Dissolveand makeup to the volume with Mobilephase.

Validation:

System Suitability Parameters: The system suitability parameters were determined by injecting standard solution of Methylcobalamin five times and the percentage RSD for Area and Retention Times to be checked. The % RSD for the area of five replicates of standard injections should not be more than 2.0.

Specificity: Checking of the interference of diluent and placebo at the retention time of Methylcobalamin in the optimized method. We should not find any interfering peaks in blank and placebo at retention times of Methylcobalamin.

Precision:

Preparation of standard solution: Weigh accurately and transfer about 10 mg of Methylcobalaminstandard into 100 ml volumetric standard flask. Dissolve and makeup to the volume with Mobilephase.

Preparation of sample solution: Weigh accurately equivalent to 10mg of sample into 100 ml volumetric standard flask. Dissolve and makeup to the volume with Mobilephase.

Linearity (forMethylcobalamin): Preparation of standard stock solution: Weighed accurately and transferred about 100mg of Methylcobalamin standard into 200 ml volumetric standardflask. 50% MethylcobalaminStandard solution: 5ml standard stock solutions was pipetted out and made up to 50ml.

75% MethylcobalaminStandard solution: 7.5ml standard stock solutions was pipetted out and made up to 50ml.

100% MethylcobalaminStandard solution: 10ml standard stock solutions was pipetted out and made up to 50ml.

125% MethylcobalaminStandard solution: 12.5mlstandard stock solutions was pipetted out and made up to50ml.

150% MethylcobalaminStandard solution: 15mlstandard stock solutions was pipetted out and made up to50ml.

Accuracy:

Preparation of Standard stock solution: Weigh accurately and transfer about 10 mg ofMethylcobalamin standard into 100 ml volumetric standard flask. Dissolveand makeup to the volume with Mobilephase.

For preparation of 80% solution: Weigh accurately and transfer powdered sample equivalent to about 8 mg of Methylcobalamin into 100 ml volumetric standard flask. Dissolve and makeup to the volume with Mobilephase. Samples were prepared in triplicates from a uniform mixture.

For preparation of 100% solution: Weigh accurately transfer powdered sample and equivalent to about 10 mg of Methylcobalamin into 100 ml volumetric standard flask. Dissolveand makeup to the volume with Mobilephase. Samples were prepared in triplicates from a uniform mixture.

For preparation of 120% solution: Weigh accurately and transfer powdered sample equivalent to about 12 mg of Methylcobalamin volumetric into 100 ml standard flask. Dissolveand makeup to the volume with Mobilephase. Samples were prepared in triplicates from a uniform mixture.

Results and Discussion:

Method development: Method development was done by changing various, mobile phase ratios, buffers etc. Mobilephase : Buffer: Acetonitrile (600:400)

Flow rate	:0.7ml/min
Column	: INERTSIL ODS C18, 3.6mm X 50 mm;3µ
Detectorwavelength	:352nm
Injectionvolume	:2µL
Runtime	:6min
Diluent	: Mobilephase
Results	: Retention time and peak shape is good.

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Figure-1: Optimized Chromatogram

Observation: Methylcobalamin was eluted at 1.193 min. and the system suitability met. so this method was optimized and to be validated.

System suitability: All the system suitability parameters were within the range and satisfactory as per ICH guidelines.

Sample ID	Met		
Sample ID	RT	AREA	
Injection -01	1.193	4331240	
Injection -02	1.195	4356897	
Injection -03	1.194	4357895	
Injection -04	1,191	4326589	

TRable-1:	System	suitability	parameter	for	Methylcobalamin

Injection -05	1.191	4421543
Average:	1.193	4358833
SD:	0.00	37875
% RSD:	0.15	0.87

Discussion: The percentage RSD of Five replicate standard solution injection was 0.15 for Retention time and 0.87 for Area which meets the requirement of less than 2. All the system suitable parameters were passed and were within the limits.

Specificity:

Table-2: Specificity for Methylcobala

	Methylcobalamin			Methylcobalamin	
Sample ID	RT	AREA			
BLANK	1.193	No peak observed			
STANDARD	1.193	4358833			
PLACEBO	1.193	No peak observed			

Discussion: Retention time of Methylcobalaminwas eluted at 1.193min. We did not find any interfering peaks in blank and placebo at retention times of Methylcobalamin in this method. So .this method was said to be specific.

Linearity:





Discussion: Five linear concentrations of Methylcobalaminwere injected. The area Vs concentration was plotted and the linearity equations obtained for Methylcobalaminy = 0.0000x-0.1035. Correlation coefficient obtained was $R^2 = 0.9999$ for the Methylcobalamin

Precision: Six independent samples were prepared and analysed as per the test method .The

percentage RSD for the content of Methylcobalamin was calculated for the samples and found 0.52 which is less than 2.0 and meets the requirements.

Accuracy: Three levels of Accuracy samples in triplicates were prepared and analysed as per the test method. The assay percentage were calculated for the each samples and found the minimum assay is 99.82 and the maximum assay percentage is 100.94 which are well within the limit of 98.0 to 102.0.

Robustness:

Robustness conditions like Flow minus (0.6ml/min), Flow plus (0.8ml/min), Wavelength

minus and Wavelength plus was maintained and samples were injected. System suitability parameters were not much affected and all the parameters were passed. %RSD was within the limit.

Parameters		Methylcobalamin	LIMIT
Linearity:		y = 0.0000x - 0.1035	r ² not less than 0.99
Regression equationY=mx+c)		(r ² =0.9999)	
Assav		100.21%	98%-102%
(% mean assay)			
Specificity		Specific	No interference of any peak
Method precision %RSD		0.73%	RSD NMT 2.0%
Intermediate precision - DAY 01 %RSD		0.52%	RSD NMT 2.0%
Intermediate precision - DAY 02 %RSD		0.55%	RSD NMT 2.0%
Accuracy %		99.82% to 100.94%	98%-102%
	FM	0.33%	
Robustness	FP	0.22%	%RSD NMT 2.0
	WM	0.63%	
	WP	0.68%	

Conclusion:

A simple, Accurate, precise method was developed for the simultaneous estimation of the Methylcobalaminin solid dosage form. Retention time of Methylcobalaminwas found to be 1.193min. and the regression equation of Methylcobalamin is y=0.0000x-0.1035 & $x^2=0.9999$.

The Method precision is 0.73%, Intermediate precision for day 01 is 052% and the intermediate precision for Day 02 is 0.55%. The assay of Methylcobalamin found as 100.21%.Since the Retention time of Methylcobalamin is less than 2.0 minutes, the developed method is cost effective with short analytical time. So the method developed was simple and economical that can be adopted in regular Quality control test in Industries.

References:

- 1. B.k Sharma, Instrumental methods of chemical analysis, Introduction to analytical chemistry, 23rd Edition Goel publication, Meerut, (2007)
- Skoog, Douglas A.; Holler, F. James; Crouch, Stanley R. (2007). Principles of Instrumental Analysis. Belmont, CA: Brooks/Cole, Thomson. p. 1. ISBN 978-0-495-01201-6
- 3. Dr.S. Ravi Shankar, Text book of Pharmaceutical analysis, Fourth edition

- 4. K. D. Tripathi, Essentials of Medical Pharmacology, 6th Edition, Jaypee brother's medical publishers (P) LTD,
- Chemistry and significance of Vitamin B₁₂ model systems by J. Halpern. D. Dolphin (Ed), B12, .1982, 1, 501-41. Wiley, N. Y.
- 6. Biosynthesis of vitamin B12 (Biosynthesis: Polyketides and Vitamins) by A.R. Battersby and F.J. Leeper, Top. Curr. Chem., 1998, **195**, 143-193.
- Organometallic chemistry in Biology: the role of Vitamin B₁₂ by J.Halpern, Bull. Soc. Chim. France, 1988, 187-191
- 8. Vitamin B₁₂. Recent discoveries cast new light on an ancient structure by A.I. Scott, Pure and Applied Chem., 1996, **68(11)**, 2057-2063
- 9. Structural and solution properties of rhodoximes: the Rh analogs of cobaloximes, a vitamin B_{12} model. by L. Randaccio, Croat. Chem. Acta, 1994, **67**(2), 235-40.
- The Discovery of Nature's pathway to Vitamin B₁₂. A 25 year Odyssey by A.I.Scott, Tetrahedron, 1994, **50** (47), 13315-13333.
- 11. How nature Synthesises Vitamin B_{12} A Survey of the Last Four Billion Years by A.I. Scott, Angew. Chem., Int. Ed. 1993, **32(9)**, 1223-1243.
- 12.A review of cobalt compounds with the chemical characteristics of vitamin B_{12} and its derivatives. Emphasis is on redox reactions,

Co-C bond formation and cleavage, and studies related to the mechanisms of B_{12} -promoted reactions.

13.Mechanism of Cobalamin-Dependent Rearrangements by B.M.Babior, Acc. Chem.Res., 1975, **8**, 376 - 384.